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# Rock-water interaction processes based on geochemical modeling and remote sensing applications in hyper-arid environment: cases from the southeastern region of Egypt

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## Abstract

**Background:** The interaction between rocks and groundwater can be considered as multi-faceted processes in hydrogeology where the groundwater chemistry can be resulted from leaching and/or minerals dissolution. The scientific approach of this study based on testing the contribution of remote sensing in providing information related to rock-groundwater interaction along with the outputs of the geochemical model. The current research aims to explore the impacts of lithologic nature and structural geology on the groundwater chemistry. The mentioned objective is significant for the future sustainable development of groundwater in hyper-arid regions. Therefore, both geology and geochemistry of water-bearing formations along the flow path to the studied aquifers were identified. The area was chosen as an example from hyper-arid region to carry out this scientific approach.

**Results:** The obtained results indicate the existence of three main aquifers: Quaternary alluvial (salinity 1253 mg/l to 18,854 mg/l), Nubian Sandstone (salinity 311 mg/l to 14,388 mg/l), and fractured basement (salinity 320 mg/l to 19,375 mg/l). The results of speciation modeling showed that studied aquifers are supersaturated with gibbsite, goethite, hematite, magnetite, aragonite, calcite, dolomite, and alunite. Nubian aquifer specifically has homogeneity results due to the similarity of water-bearing and geology of watersheds (ferruginous sandstone and ironstone). Remote sensing data and digital elevation model analyses were used for generation of the thematic layers which is affecting the groundwater occurrences and quality such as geology, geomorphology, structural lineaments, slope, flow direction, drainage lines, and elevations.

**Conclusions:** The current research concluded that the groundwater chemistry can be a signature of the rock-water interaction. The potential implications of the research are represented through a priority map for groundwater exploration for both hydro-geophysical investigation and testing wells in order to evaluate the groundwater capabilities where the data from the recorded wells were used to calibrate this map.

**Keywords:** Rock-water interaction, Hydro-geochemistry, Groundwater, Geology, Remote sensing, GIS

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## Background

Water scarcity in the hyper-arid region, such as the south-eastern Egypt, represents the main challenge to the hydro-geologists where they should work to evaluate and protect any source of water. The deficiency of water resources in these hyper-arid regions could affect different activities (including economics and humans); therefore, numerous efforts are essential to avoid any of the expected disasters and conflicts which might occur as a result of limited water resources (Gheith and Sultan 2002). The decision-makers in Egypt have an important challenge to manage water resource for sustainable development (Ahmed and Abdelmohsen 2018). In this regard, the consideration of the processes which affect groundwater and its chemistry through rocks interactions becomes critical and necessary. Water lack is the main problem affecting the sustainable development plans in the southeastern region of Egypt where the water delivered to the Red Sea Governorate is either from the Nile River through pipes or depending on seawater desalinization (Abdel Moneim 2005). The precipitation is negligible over the South Eastern Desert without any source of recharge (e.g., Lake Nasser and Northern Sudan Platform) (Ahmed and Abdelmohsen 2018). Accordingly, groundwater is considered as an alternative solution to face these problems, and therefore, the area attracted many authors such as Abdel Moneim (2005), Yousef et al. (2009), Shawky et al. (2012), Soussa et al. (2012), and Khalil (2014) to investigate geology and water resources. The groundwater contains different types of minerals carried in solution where the types and concentration of these minerals are mainly conditional on multi-factors such as rocks chemistry, surface and groundwater interactions, the geologic setting of the preferred pathways for groundwater flow, and any other sources for water pollution. The groundwater chemistry represents the final product of the interaction between water and rocks belonging to different geological times. These interpret why groundwater contains varieties of many dissolved chemical constituents with different types and concentrations. The main objective of the current research is to explore the impacts of lithologic nature and structural geology on the groundwater chemistry. The mentioned objective is significant for the future sustainable development of groundwater in hyper-arid regions. In this context, the research has been carried out to investigate the processes resulted from the interaction between different rocks and groundwater. In this manuscript, rock terms are expressing about two types of meaning: the water-bearing formation and the rocks which formed the watersheds that affect the groundwater. The presented scientific approach of this study is based on exploring the integration of remote sensing and satellite applications in providing spatial information related to rock-water interaction processes along with the outputs and results of the geochemical model.

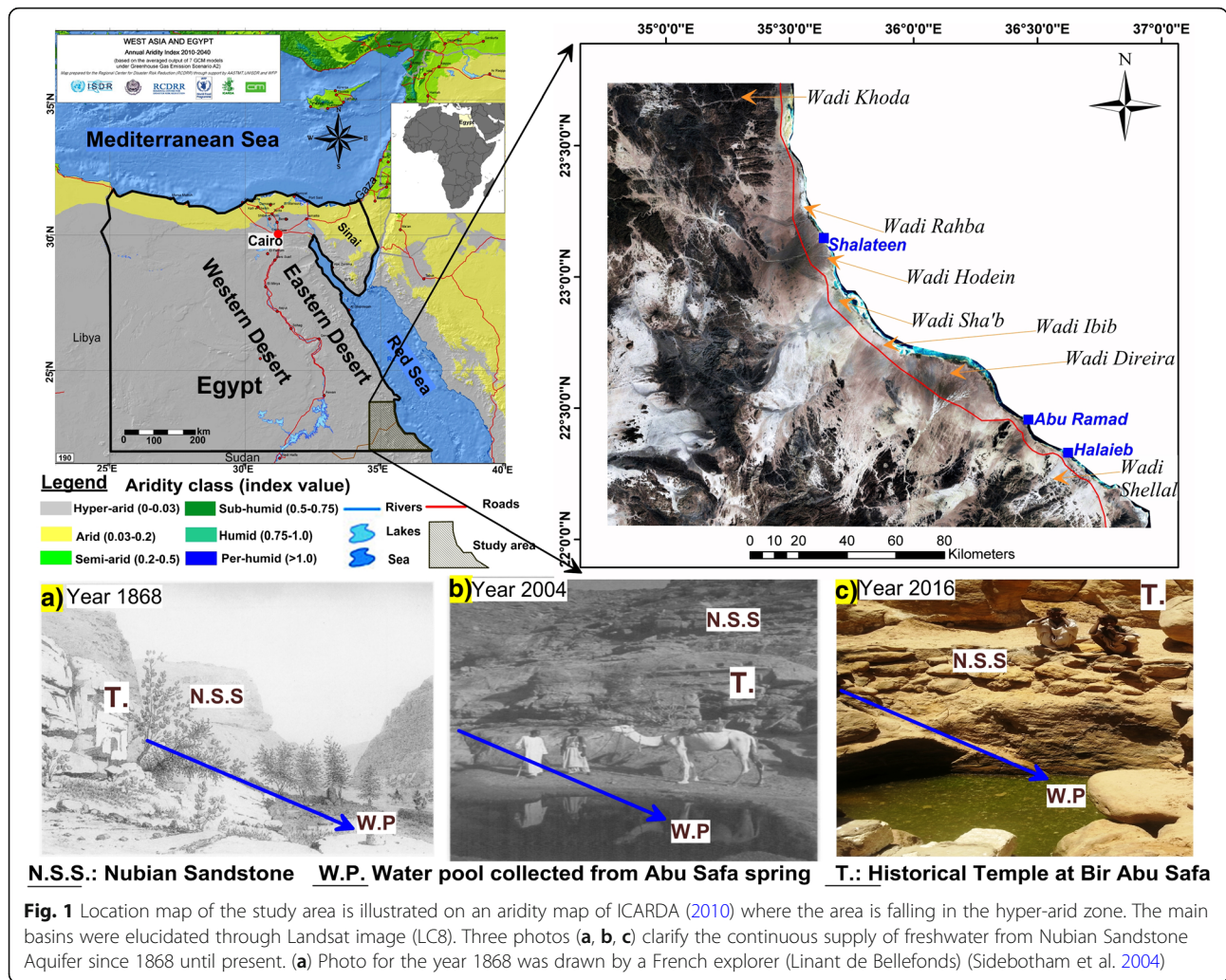
## Site descriptions

The study area is falling in the hyper-arid zone of Egypt according to the aridity index map (ICARDA 2010) which reveals the shortage of water resources supplied to this area. This is compatible with Gravity Recovery and Climate Experiment (GRACE) data which proved high groundwater depletion rates over the southern parts of Egypt and Libya (Mohamed et al. 2017). The southeastern region of Egypt represents one of the promising regions which is not subjected to intensive development projects yet due to the scarcity of water or inability to use. The area under investigation covers about 22,000 km<sup>2</sup> and is located along the Red Sea coast as a part of the Eastern Desert of Egypt (Fig. 1). The study area lies between latitude 22° N and 23° 40' N and longitude 34° 50' E and 37° E. The temperature reaches its maximum in the summer and ranges between 45 and 50 °C (Ageeb et al. 2007). The rainfall plays an important role as it is considered the main source for the aquifer recharge. The average annual rainfall ranges from 3 to 50 mm/year (Embaby et al. 2016), while relative humidity ranges between 28 and 52% (Hassanein et al. 2004).

## Geomorphology

The study area can be classified into the following geomorphic unit, as shown in (Fig. 2):

- Coastal plain—it extends westward of the Red Sea shoreline for a distance ranging from 5 to 25 km. The plain is covered by sand sheets, alluvial, and deltaic deposits. Along the shoreline, the plain is characterized by the presence of sabkha patches. It is partly covered by bushes and scrubs.
- Red Sea terrains—it borders the coastal plain and partly the Piedmont Plain. The Red Sea terrains are covered by the mountains oriented in a northwest-southeast direction parallel to the Red Sea shoreline. The elevation is variable from 388 to 1419 m above sea level. The slope ranges from 15 to 30 m/km due east.
- Isolated hills—few isolated hills are spreading over the study area, such as Gabal Umm Rasin and Gebel Elba. These hills are composed mainly of Precambrian basement rocks of different types.
- Sedimentary blocks—these blocks are located in a limited area in the west side where it is composed of Cretaceous rocks belonging to Nubian Sandstone.
- Piedmont Plain—this unit is controlled by the action of drainage basins which dissected the Red Sea terrains and attains to expand the distance between each of the coastal plain and Red Sea terrains. This unit is mainly covered by alluvial deposits and natural vegetation.



**Fig. 1** Location map of the study area is illustrated on an aridity map of ICARDA (2010) where the area is falling in the hyper-arid zone. The main basins were elucidated through Landsat image (LC8). Three photos (a, b, c) clarify the continuous supply of freshwater from Nubian Sandstone Aquifer since 1868 until present. (a) Photo for the year 1868 was drawn by a French explorer (Linant de Bellefonds) (Sidebotham et al. 2004)

- Wadies and alluvial fans—the area under consideration is dissected by about nine main wadies which drain their surface water to the Red Sea eastwards. Most of these wadies are of small areas and short lengths.

**Geology of the study area**

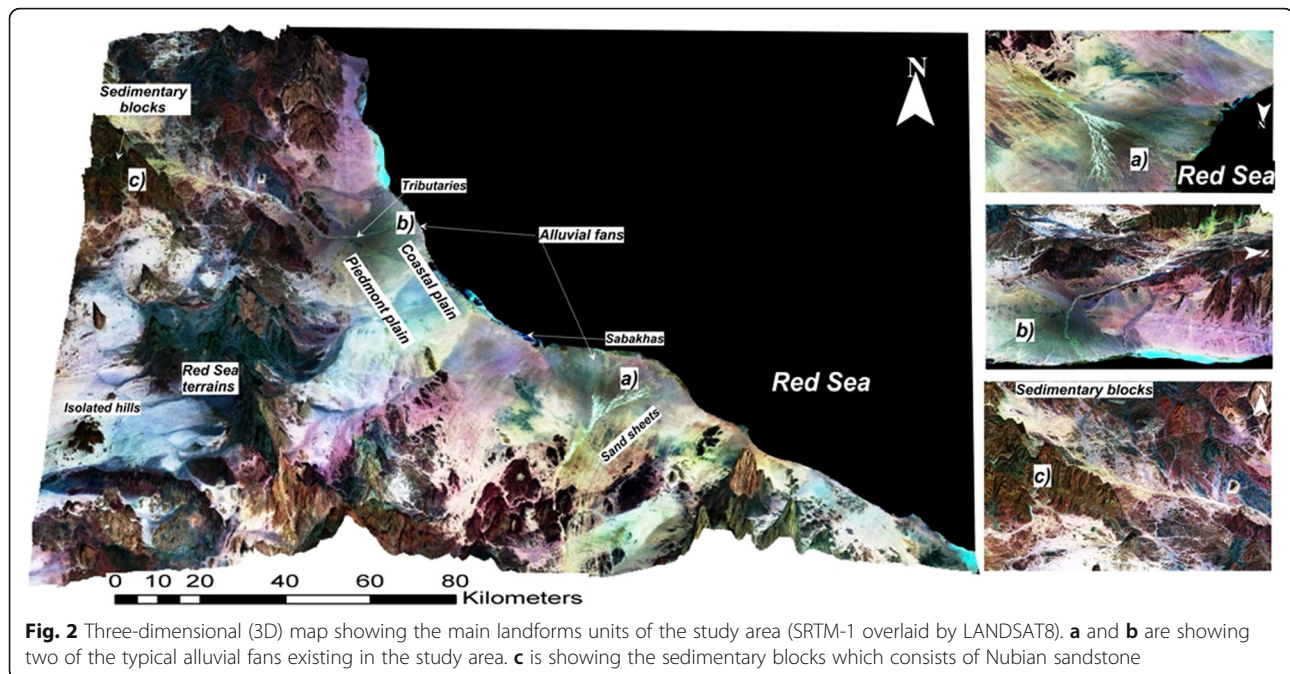
To the west of the coastal zone, the area is covered completely (except the area of sedimentary blocks) by basement rocks, which consist mainly of granites, gneisses, schists, and quartzite of Precambrian age (Fig. 3). These basement rocks are overlain non-conformably by about 430 m of thick sandstone succession of Nubian facies (Elewa (2000). The sandstones consist of three formations, namely Abu-Agag, Timsah, and Umm Barmeil formations at the top. Towards the southwest, the sandstone in general dips due east and southeast. Isolated patches of Miocene sediments are located as rare exposures which consist mainly of alternating limestone and marl. Quaternary sediments comprising wadi fills and alluvial wadi deposits cover a vast area. Most of the wadies are filled with

alluvial deposits with a thickness that varies between few meters and about 20 m. Generally, the thickness increases from west to east. Numerous traverse faults have NW-SE and E-W direction. On the other hand, a small group of minor faults exists in a NE-SW direction.

**Methods**

**Fieldwork**

During fieldwork, the geomorphologic and geologic settings of the study area were determined. Thirty sites of the study area were investigated in detail with regard to geology where the surface outcrops were identified. A total of 75 rock samples from different geological units were collected and described. The field data were used for an area-wide interpretation based on Landsat satellite images and geologic map (Conoco 1987). Also, a field trip was achieved during the summer of 2016. Though this fieldwork, the groundwater samples have been collected (37 groundwater samples representing three aquifers). The site locations have been acquired



using Global Positioning System (GPS), while other measurements including reduction potential (redox potential, oxidation/reduction potential, “Eh”), the hydrogen ion concentration (pH), and the electrical conductivity (EC) were achieved in situ. Also, the hydrological data including depth to water table, total well depth, and water-bearing formations, as well as corresponding geomorphologic units, were also determined. The main factors which affect the groundwater quality and occurring were elucidated in the site where each water sample has a special case which subjected to investigation.

### Chemical analyses

The total of 37 groundwater samples was analyzed at the water analysis unit of the central laboratory of Desert Research Center (DRC, Cairo, Egypt) according to ASTM 2002. The chemical analyses of the collected water samples included the determination of EC, total dissolved salts (TDS), pH, and concentrations of major cations ( $\text{Na}^+$ ,  $\text{K}^+$ ,  $\text{Mg}^{2+}$ ,  $\text{Ca}^{2+}$ ) and major anions ( $\text{Cl}^-$ ,  $\text{SO}_4^{2-}$ ,  $\text{CO}_3^{2-}$ ,  $\text{HCO}_3^-$ ) by ion chromatography (ICS-1100, Dionex, Sunnyvale, CA, USA). Also, the trace element (Al, B, Cd, Co, Cr, Cu, Fe, Mn, Mo, Ni, Pb, Ti, Sr, P, V, and Zn) concentrations of the water samples were determined using inductively coupled argon plasma (ICAP, Thermo 6500).

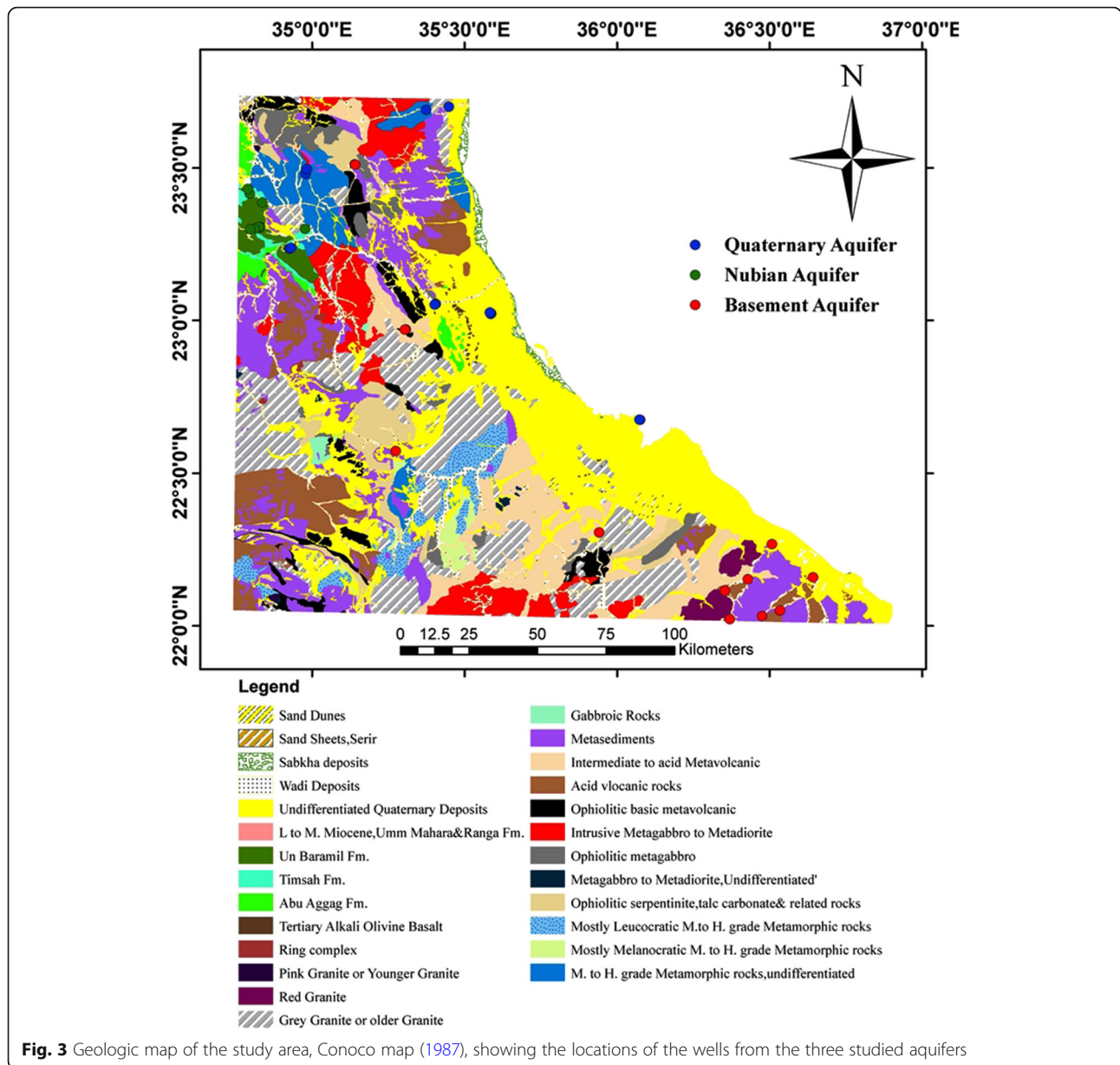
### Geochemical modeling and rocks geochemistry

The obtained chemical analyses which included the concentrations of both major and trace elements are

representing the input data for the geochemical modeling of the PHREEQC code (Parkhurst and Appelo 1999; USGS). The PHREEQC code is used to calculate the saturation indices (SI) for minerals of interest, where this index determines whether a solution has a thermodynamic potential for further dissolution or precipitation of a mineral. The saturation indices (SI) of the relevant mineral are determined through the equation  $\text{SI} = \log(\text{IAP}/\text{KT})$ , where IAP is the ion activity products and KT is the solubility product of a mineral for a given temperature. The other parameters such as Eh and pH were also included in the database before the model running. The outputs of the model were tabulated where we focused on the minerals that saturated the studied groundwater. On the other hand, the geochemistry of the water-bearing formation and watersheds in the study area were obtained from different sources which include the works of Hassan (2015), Abd El-Naby and Frisch (2006), Khalil (1997), Farouk et al. (2015), and Shahin et al. (2016).

### GIS and remote sensing

The digital elevation model (DEM) of the Shuttle Radar Topography Mission (SRTM-C) and the Enhanced Thematic Mapper Plus (ETM+) images were used to obtain data for the current research. The obtained data from the satellite images was employed for the visual interpretation of surface features and geology as well as the structural lineaments. All DEM data and satellite images were projected to the Universal Transverse Mercator (UTM) with the WGS84 datum. After processing of SRTM-C data by constructing a mosaic to extract the



**Fig. 3** Geologic map of the study area, Conoco map (1987), showing the locations of the wells from the three studied aquifers

watersheds, the hydrology tools of ESRI Arc GIS v.10.4 were used to identify and fill all sinks in the resulting mosaic. Many GIS data layers were obtained through the analyses of DEM including watersheds, slope, flow direction, drainage networks, elevation, and aspect as well as the three-dimensional layer (3D) to illustrate the main landforms as well as geological and geomorphological units. Landsat GeoCover ETM+ (2000) with a spatial resolution of 14.5 m (band combination 742) was used with the geological map of (Conoco 1987) to determine the geology of the study area. Structural lineaments were extracted through band combination (741 and 754 of the ETM+) where the edge detection filtering in ERDAS IMAGIN 9.3 depends on the variation of digital pixel

values. The geological map was also used to elucidate the lineaments structures (faults and/or fractures).

### Results

Results are presented according to the before-mentioned workflow.

#### Groundwater occurrences

In the current study, three aquifers were chosen to be subjected for investigation where they are providing some suitable groundwater (Table 1). These aquifers are from young to old as follows:

**Table 1** Hydrogeological data of the wells tapping the studied aquifers

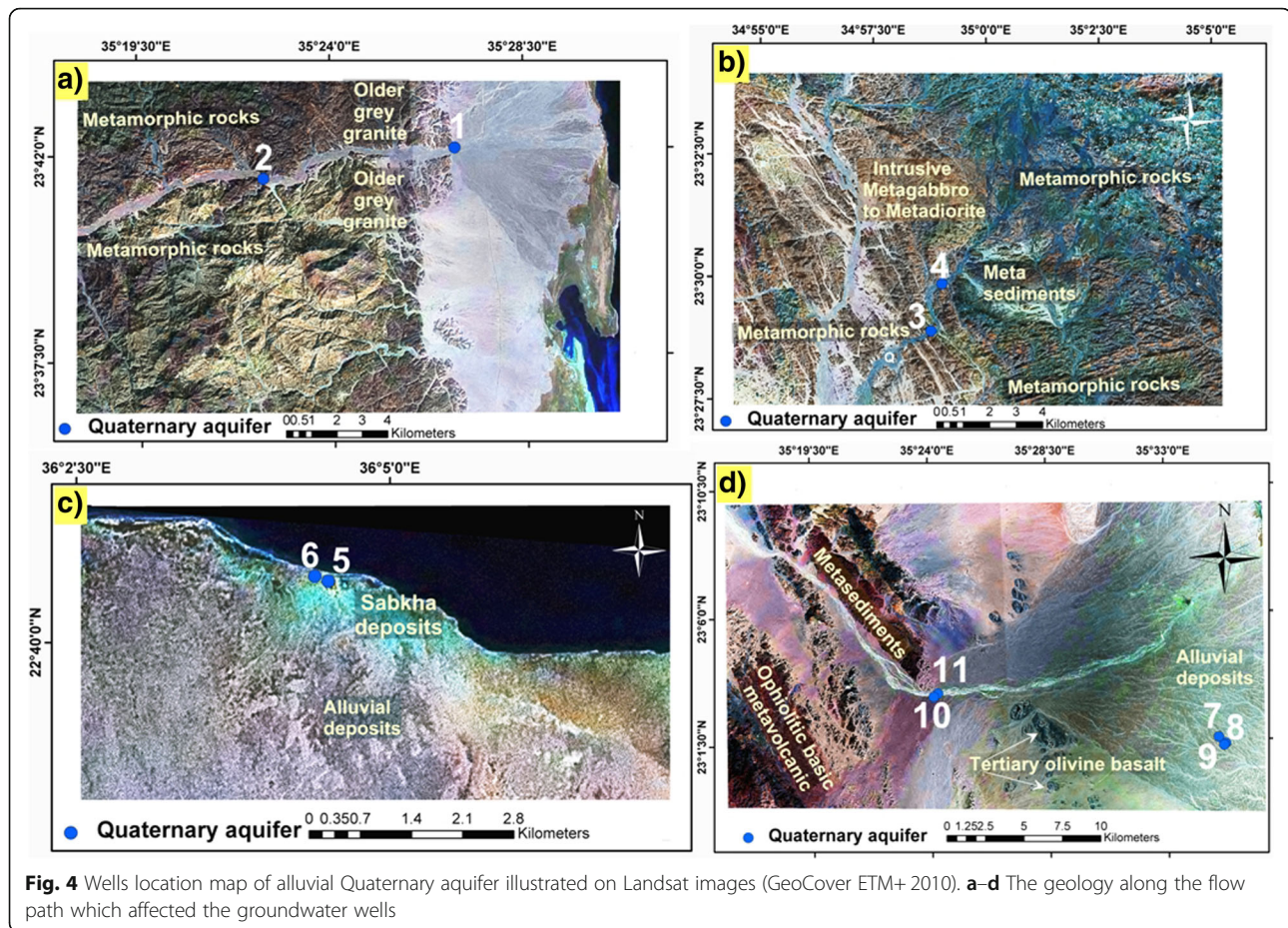
Well no.	Long. (E)	Lat. (N)	Aquifer		Total Depth (m)	Depth to water (m)	TDS (mg/l)
			Formation	Type of drilling			
1	35.448	23.701	Alluvial	Hand-dug	20	12.35	2427
2	35.374	23.691		Hand-dug	20	14	2153
3	34.978	23.481		Hand-dug	–	6.53	2068
4	34.983	23.496		Hand-dug	15	10.3	1253
5	36.076	22.675		Hand-dug	2	1.35	8386
6	36.074	22.675		Hand-dug	2	1.05	9750
7	35.587	23.023		Hand-dug	25	20.3	11,053
8	35.586	23.022		Hand-dug	25	20.2	18,854
9	35.583	23.026		Hand-dug	–	20	10,095
10	35.405	23.054		Hand-dug	30	23.4	7880
11	35.402	23.053		Hand-dug	30	23.3	6446
12	34.926	23.236	Nubian Sandstone	Drilled	120	24.5	776
13	34.931	23.239		Drilled	150	25	14,388
14	34.830	23.307		Drilled	87	26.9	757
15	34.820	23.300		Drilled	90	24.7	778
16	34.825	23.320		Drilled	310	25.93	915
17	34.811	23.301		Drilled	93	28	943
18	34.976	23.301		Spring	Flowing		341
19	34.796	23.300	Spring			425	
20	34.837	23.386	Spring			356	
21	34.800	23.400	Spring			311	
22	34.795	23.416	Spring			346	
23	34.789	23.431	Spring			579	
24	35.140	23.512	Fractured basement	Hand-dug	9.5	4.7	320
25	35.141	23.511		Hand-dug	10	5.17	353
26	35.941	22.306		Hand-dug	29	28.35	3522
27	35.274	22.572		Hand-dug	17	14.5	502
28	36.643	22.159		Hand-dug	24	20.3	19,375
29	36.536	22.051		Hand-dug	22	18.09	817
30	36.534	22.050		Hand-dug	15	10.6	1139
31	36.475	22.032		Hand-dug	30	27	1027
32	36.369	22.022		Hand-dug	33	29.5	4289
33	36.353	22.115		Hand-dug	29	26	2079
34	36.429	22.153		Hand-dug	25	22	4986
35	36.508	22.267		Hand-dug	12	8	6151
36	35.307	22.971		Hand-dug	4.5	1.4	5138
37	35.304	22.970		Hand-dug	3.5	1.4	13,673

**Quaternary aquifer**

The aquifer is composed of sands, mudstone, and different sizes of gravels (granules, pebbles, and boulders) derived from the basement rocks especially that located along the surface water path and in the vicinities of the wells (Fig. 4). This aquifer is mainly recharged from the

occasional runoff water of the intermittent streams; its deposits belong to the Pleistocene and the recent. This aquifer is covering most of the wadi floor with alluvial deposits, which can be subdivided into two groups:

The first group is the wadi fillings in the depressions over alluvial fans and along alluvial courses. These sediments



have a variable thickness which occurred overlies sedimentary and basement rocks. The second group is coral and beach sand deposits where the alluvial sediments are of special hydrogeologic significance. It is covered by sand all over the area and sometimes by silt and gravel. The Quaternary aquifer is explored through 11 wells where the total depth ranges from 2 to 30 m, and the depth to water is varying between 1.05 and 23.4 m (Table 1).

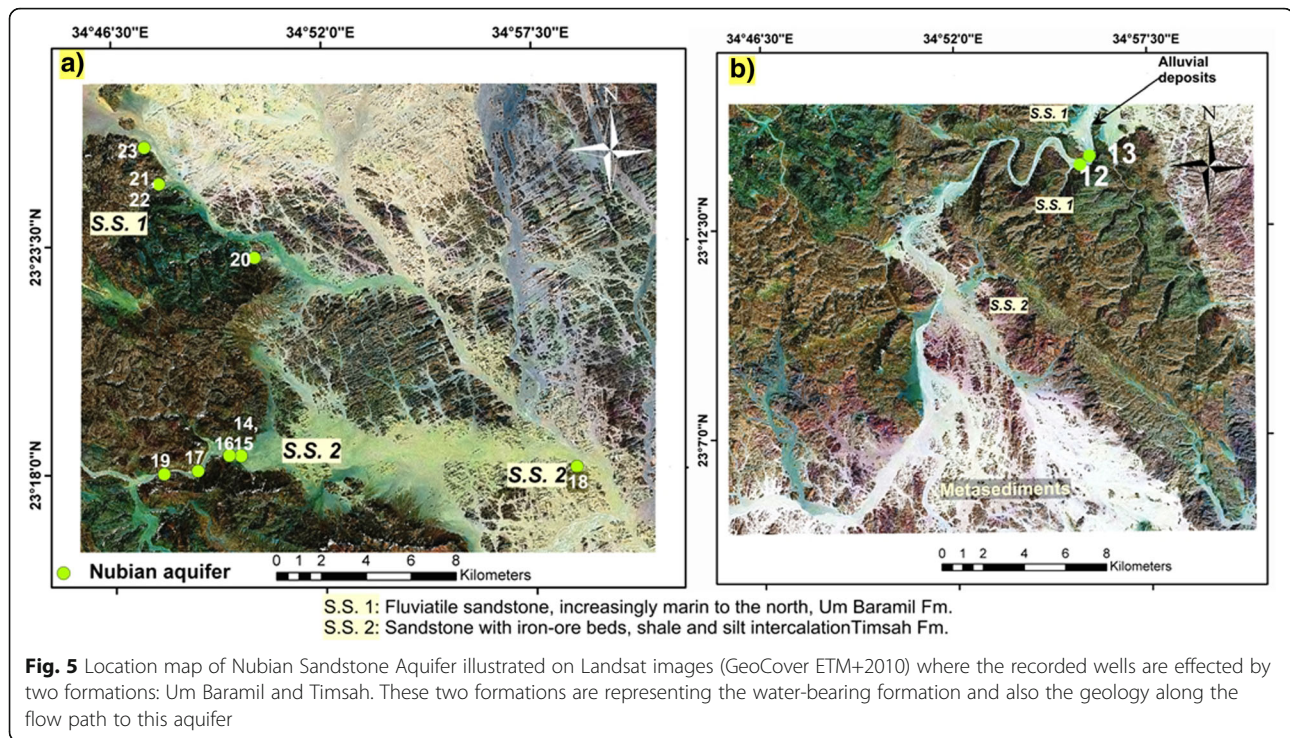
#### **Nubian Sandstone Aquifer of Upper Cretaceous ages**

The Nubian Sandstone Aquifer occupies mostly an area of Wadi Hodein Basin; in such rocks, faults and slight folding were associated with the Red Sea Rifting blocks. These blocks act as independent aquifers, which flowing naturally in the form of springs (six springs) with the discharge rates up to 21 m<sup>3</sup>/h. Most of the groundwater of the Nubian Sandstone is slightly fresh. The aquifer consists of sandstone which is fine to coarse-grained with clay and siltstone intercalation (Fig. 5). On the other hand, six wells were drilled penetrating the Nubian Sandstone Aquifer up to 310 m, while the depth to water ranges between 24.5 and 28 m. The hydrologic parameters of this aquifer were deduced from pumping tests carried out by Elewa (2000). He stated that the

transmissivity ranges from 2.72 to 72.4 m<sup>2</sup>/day, and the recharge processes are mainly due to the direct infiltration from rainfall and partly from surface runoff through the fractured sandstone. This is confirmed through continuous discharge from the springs (since 1868).

#### **Fractured basement rock of Precambrian age**

This aquifer comprises the fractured Precambrian basement which represents a part of the Red Sea mountainous area. These rocks include granites (old and young), granitoids, gneisses, migmatites, schists, metasediments, diorites, and quartzites. These rocks are characterized by high and dense fractures, and faults intersecting with each other facilitate the recharge process. This structural setting provides chances for groundwater to be accumulated and moved through the fracture systems. The occurrence of groundwater depends mainly upon the depth, width, and extension of fractures, where the water quantity is frequently limited but provides a water supply for the local population. The aquifer is penetrated through 14 hand-dug wells (Fig. 6) where the total depth ranges between 3.5 and 33 m while the depth to water varies between 1.4 and 29.5 m (Table 1).



### Chemistry of groundwater

The data of analyses of the groundwater wells are shown both in Tables 2 and 3. The groundwater salinity in the study area ranged from fresh to brackish and saline. The lowest total dissolved salt (TDS) value (311 mg/l) was recorded in well no. 21 of the Nubian Sandstone Aquifer, while the highest value was obtained from well no. 28 of the fractured basement aquifer. The alluvial aquifer showed variation in the salinity values that ranged between 1253 and 18,854 mg/l. The majority of groundwater samples of this aquifer have one order of ion dominance:  $\text{Na} > \text{Ca} > \text{Mg}$  with  $\text{Cl} > \text{SO}_4 > \text{HCO}_3$ . The analyses of groundwater belonging to the Nubian Sandstone Aquifer reveal relative symmetry in their chemical composition. All groundwater samples were freshwater (except one sample of well no. 13, with TDS of 14,388 mg/l). The TDS values ranged between 311 and 943 mg/l. The ion dominance of this aquifer includes two groups:  $\text{Na} > \text{Ca} > \text{Mg}$  with  $\text{Cl} > \text{HCO}_3 > \text{SO}_4$  which was recorded in the most of groundwater samples and  $\text{Na} > \text{Mg} > \text{Ca}$  with  $\text{Cl} > \text{SO}_4 > \text{HCO}_3$  in few wells.

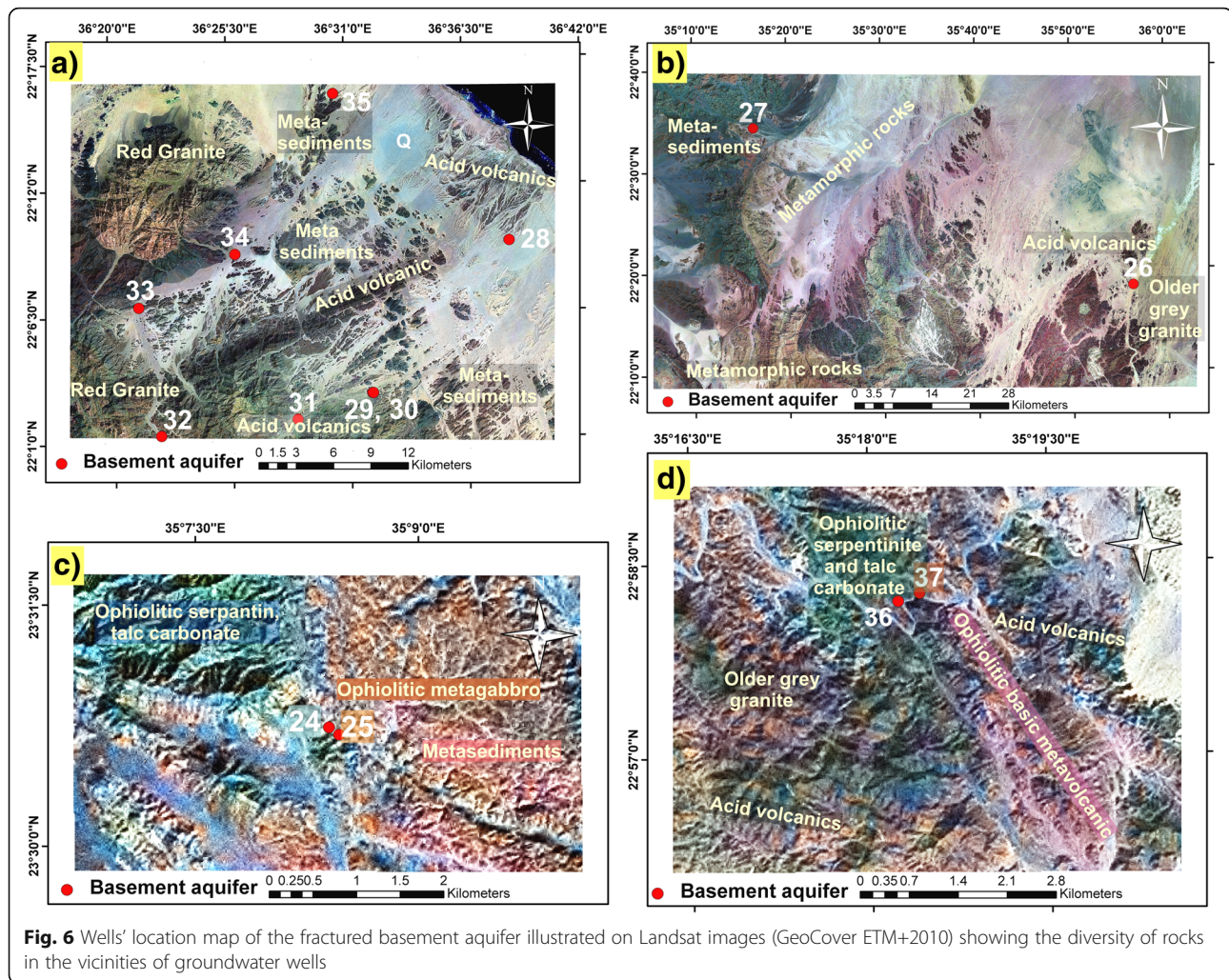
On the other hand, the chemical results of the groundwater samples tapped from the fractured basement aquifer showed that a wide range of salinity and the water samples can be classified from fresh to brackish, and only two wells were classified as saline (well nos. 28 and 37). According to the values of TDS, it had a wide range; it ranged from 330 to 19,375 mg/l. Three groups of ion dominance were observed:  $\text{Na} > \text{Ca} > \text{Mg}$  with  $\text{Cl} > \text{SO}_4 > \text{HCO}_3$  which was noted in the most samples,  $\text{Na} > \text{Mg} >$

$\text{Ca}$  with  $\text{Cl} > \text{SO}_4 > \text{HCO}_3$  which was recorded in three groundwater samples, and finally a group that was characterized by the freshwater where bicarbonates are greater than sulfates  $\text{HCO}_3 > \text{Cl} > \text{SO}_4$  with cation dominance varies between  $\text{Na} > \text{Mg} > \text{Ca}$  and  $\text{Ca} > \text{Mg}$ .

The excess of  $\text{Cl}^-$  and  $\text{Na}^+$  can be explained by the combined effect of the unfractionated aerosols belonging to marine environment and dissolution of watershed rocks. The marine aerosols are most likely a dominant source of solutes where marine aerosols may be deposited as dry salts on the land surface and periodically dissolved and transported downward towards the aquifer during sporadic rainfall events. Some of these solutes may show high concentration through evaporation processes (such as, halite, gypsum, and other salts) where these salts may be accumulated for many years and then transported to the underground aquifer when a significant rainfall event occurs. On the other hand, the alkaline earth elements may be derived by dissolution of silicate minerals (e.g., plagioclase feldspar, chlorite, or biotite), carbonates (dolomite or calcite), gypsum, and/or cation exchange of  $\text{Na}^+$  for  $\text{Ca}^{2+}$ , and  $\text{Mg}^{2+}$  on clay minerals. The excess of  $\text{Ca}^{2+}$  and  $\text{Mg}^{2+}$  does not result from the dissolution of carbonate rocks (calcite, dolomite) where the distribution is limited. Therefore, it can be attributed to metavolcanic and basic rocks.

Trace concentrations for 15 elements of the studied groundwater samples are generally between 0.0001 and 22.82 mg/l (Table 3). Concentrations of Fe are Mn are relatively low in the studied groundwater except for the





water from Nubian Sandstone Aquifer that showed a noticeable increase and ranging between 0.08 and 10.49 mg/l for Fe and between 0.02 and 2.83 mg/l for Mn. This is due to the lithologic nature of Nubian sandstones which are rich in iron and manganese oxides. On the other hand, Sr and Al show notable recordings in the three aquifers where higher concentrations are present in the alluvial aquifer (19.5 mg/l, 20.46 mg/l, and 22.8 mg/l for Sr and 6.67 mg/l for Al). This is due to the presence of Sr and Al sources which are found both in the water-bearing rocks and also along the surface water path that recharges the groundwater. These sources are represented in the basement rocks, sandstone sediments, and carbonate rocks. Other trace elements did not show any constant trend as a result of a great diversity in the rocky nature for each of aquifers and watersheds.

#### Hypothetical salts of groundwater

Regarding the hypothetical salt combinations in the three investigated aquifers of the studied area, there are

six main assemblages detected in the groundwater that are summarized in (Table 4). Generally, it is obvious that the groundwater samples of the study area are characterized by five different assemblages (II–VI); this indicates that there is no connection between them and affected by different conditions. The groundwater samples of the alluvial aquifer possessed three of the hypothetical salts assemblages, from IV to VI. Also, the groundwater samples of both the Nubian Sandstone and the fractured basement aquifers are characterized by three hypothetical salts assemblages II, III, and V for the Nubian Sandstone and III, IV, and VI for the fractured basement aquifer. Generally, the groundwater is characterized by meteoric water origin which is affected by leaching and dissolution processes. The presence of  $\text{Na}_2\text{SO}_4$ ,  $\text{MgSO}_4$ , and  $\text{Mg}(\text{HCO}_3)_2$  salts reflects the effect of dissolution with leaching processes of earthly salts as the groundwater is characterized by assemblage II. The groundwater that is characterized by assemblages (III, IV, V and VI) that contain  $\text{MgCl}_2$  salt means that this groundwater

**Table 2** Chemical data (major elements) of the investigated groundwater samples

Well no.	Aquifer	Eh	pH	EC ( $\mu\text{S}/\text{cm}$ )	TDS ( $\text{mg}/\text{l}$ )	Cation ( $\text{mg}/\text{l}$ )				Anion ( $\text{mg}/\text{l}$ )				SAR
						$\text{Ca}^{2+}$	$\text{Mg}^{2+}$	$\text{Na}^+$	$\text{K}^+$	$\text{CO}_3^{2-}$	$\text{HCO}_3^-$	$\text{SO}_4^{2-}$	$\text{Cl}^-$	
1	Alluvial	127	7.7	4200	2427	196.30	111.6	500	7	0.00	137.25	464.06	1079.4	7
2		129	7.6	3970	2153	242.60	108.0	360	6	0.00	134.20	469.5	899.50	4.8
3		82	8.4	3420	2068	135.63	100.6	410	14	30.00	219.60	780	488.30	6.5
4		81	8.4	2408	1253	58.56	114.0	210	13	30.00	259.25	388.92	308.40	3.6
5		104	7.6	13,810	8386	654.40	318.4	1850	20	15.00	79.30	2044.65	3443.8	14
6		107	7.9	16,410	9750	703.60	390.9	2200	17	15.00	122.00	1942.35	4420.4	16.5
7		119	7.4	19,940	11,053	1280.0	555.7	2000	16	0.00	54.90	800	6373.6	11.7
8		124	7.6	33,100	18,854	2117.0	778.8	3700	31	0.00	61.00	1274	10,922	17.4
9		119	7.4	17,320	10,095	1375.0	373.2	1750	20	0.00	45.75	900	5654	10.8
10		100	7.4	13,340	7880	1075.0	190.3	1400	16	0.00	67.10	1567.6	3598	10.3
11		90	7.5	11,550	6446	1100.0	127.7	1050	17	0.00	45.75	633.05	3495	8
12	Nubian Sandstone	130	7.8	1394	776	80.69	40.46	125	10	0.00	167.75	198.47	237.47	2.8
13		145	7.3	25,350	14,388	562.90	727.3	3700	36	15.00	207.40	1841.9	7401	24
14		94	8.5	1621	757	15.63	11.69	245	7	0.00	85.40	64.72	370	11.5
15		97	8.1	1653	778	21.39	36.09	200	9	0.00	115.90	155.45	298.12	6
16		98	8	2045	915	27.80	47.46	230	11	15.00	152.50	178.31	328.96	6
17		96	7.9	2186	943	28.00	58.39	230	11	15.00	146.40	146	380.36	5.6
18		93	7.4	703	341	59.00	20.03	32	6	0.00	122.00	52	110.51	1.1
19		96	7.2	862	425	76.56	22.79	40	9	0.00	158.60	61.192	136.21	1.1
20		93	8.3	723	356	42.46	26.99	40	12	15.00	158.60	53.152	87.38	1.18
21		96	8	655	311	49.76	20.11	32	7	15.00	122.00	37.884	88.67	1.1
22		88	7.9	696	346	58.32	20.84	32	6	15.00	122.00	50.196	102.80	1.1
23		94	7.2	1003	579	99.43	25.03	65	7	0.00	167.75	144.615	154.20	1.51
24	Fractured basement	80	8.3	685	320	13.95	47.80	36	5	15.00	170.80	30.3	86.10	1.04
25		82	8.5	768	353	13.85	50.31	44	7	15.00	183.00	28.636	102.80	1.23
26		90	8.1	5280	3522	140.10	82.84	620	4	15.00	122.00	1751.34	848.10	10
27		75	8.2	862	502	15.91	68.50	55	3	15.00	122.00	178.4	105.37	1.33
28		133	6.7	35,800	19,375	3839.0	477.2	2450	25	0.00	39.65	228.12	12,336	9.9
29		66	8.1	1837	817	45.69	19.34	230	2	18.00	195.20	105.94	298.12	7.19
30		68	8.1	2068	1139	63.72	20.34	320	3	15.00	207.40	243.1	370.08	8.94
31		80	7.8	1931	1027	159.80	40.53	160	3	15.00	213.50	120	421.48	2.93
32		97	7.4	7240	4289	621.00	178.3	620	2	15.00	106.75	1000	1799	5.62
33		85	7.4	3900	2079	294.60	82.86	320	2	18.00	131.15	396.9	899.50	4.24
34		93	7.2	8290	4986	678.00	194.1	750	3	0.00	106.75	1200	2107	6.5
35		94	7.3	10,500	6151	737.50	171.1	1100	10	15.00	134.20	1994.15	2056	9.4
36		85	7.8	9210	5138	487.90	164.9	1100	16	6.00	97.60	1103.75	2210	11
37		96	7.6	23,340	13,673	1200.0	503.0	2800	20	0.00	122.00	3331.9	5756.8	17
Rainfall		–	7.0	125	80	10.80	3.01	15	2	0.00	49.00	10	15.00	1.00
Sea water		124	8	66,400	38,021	721.20	1668	11,000	330	21.00	64.05	1761.5	22,487	51.4
P.L			6.5–8.5		1000	200	150	200	–	–	–	250	500	

P.L permissible limit according to EHCW (2007), SAR sodium adsorption ratio

**Table 3** Minor and trace elements concentration (in mg/l) of the investigated groundwater samples

Well no.	Aquifer	Al	B	Ba	Cr	Cu	Fe	Mn	Mo	Ni	Pb	Sr	Ti	V	Zn	P
1	Alluvial	0.017	0.716	0.024	0.019	0.000	0.034	0.013	0.035	0.003	0.054	3.944	0.049	0.048	0.017	0.050
2		6.767	0.626	0.123	0.013	0.016	3.940	0.417	0.007	0.028	0.065	3.819	0.050	0.027	0.064	0.582
3		0.119	1.090	0.033	0.013	0.000	0.040	0.006	0.023	0.006	0.018	0.990	0.008	0.023	0.000	0.008
4		0.000	0.675	0.045	0.008	0.000	0.000	0.000	0.034	0.002	0.015	1.022	0.002	0.043	0.000	0.039
5		0.108	2.607	0.028	0.021	0.000	0.109	0.011	0.056	0.006	0.008	9.823	0.032	0.022	0.012	0.113
6		0.107	3.069	0.033	0.014	0.004	0.099	0.013	0.048	0.006	0.008	10.840	0.027	0.027	0.003	0.217
7		0.785	2.574	0.051	0.017	0.011	1.541	0.066	0.006	0.009	0.002	20.460	0.058	0.017	0.223	0.037
8		0.477	3.231	0.067	0.030	0.000	0.739	0.019	0.007	0.008	0.003	22.820	0.065	0.026	0.033	0.018
9		0.269	1.807	0.036	0.020	0.000	0.479	0.023	0.006	0.007	0.010	19.560	0.049	0.029	0.043	0.018
10		0.032	0.779	0.021	0.009	0.001	0.000	0.224	0.005	0.005	0.000	13.190	0.028	0.015	0.000	0.004
11		0.035	0.949	0.032	0.013	0.000	0.000	0.094	0.017	0.006	0.004	15.610	0.032	0.023	0.009	0.000
12	Nubian Sandstone	0.000	0.161	0.039	0.023	0.396	1.680	1.255	0.000	0.006	0.022	0.934	0.014	0.000	0.261	0.004
13		1.213	2.716	0.021	0.011	0.002	10.490	2.831	0.000	0.015	0.020	16.040	0.087	0.000	9.551	0.040
14		0.180	0.073	0.070	0.024	0.000	1.405	0.052	0.000	0.002	0.030	0.663	0.000	0.000	3.067	0.055
15		0.249	0.099	0.030	0.012	0.007	4.289	0.215	0.000	0.002	0.046	0.234	0.002	0.004	1.915	0.062
16		0.043	0.134	0.016	0.017	0.002	4.153	0.197	0.000	0.002	0.026	0.412	0.000	0.000	1.162	0.018
17		0.045	0.149	0.028	0.016	0.001	0.246	0.009	0.000	0.006	0.015	1.302	0.015	0.004	0.078	0.060
18		0.000	0.000	0.047	0.026	0.000	0.635	0.540	0.000	0.004	0.010	0.363	0.008	0.000	0.011	0.000
19		0.045	0.000	0.112	0.004	0.000	0.366	1.662	0.000	0.003	0.011	0.453	0.017	0.000	0.013	0.153
20		0.040	0.000	0.064	0.013	0.000	0.366	0.024	0.004	0.000	0.011	0.381	0.004	0.000	0.000	0.000
21		0.044	0.000	0.070	0.004	0.007	0.108	0.045	0.000	0.000	0.010	0.312	0.010	0.000	0.002	0.000
22		0.079	0.000	0.017	0.016	0.000	0.083	0.090	0.000	0.001	0.010	0.385	0.016	0.000	0.007	0.024
23		0.040	0.000	0.068	0.012	0.000	0.137	0.143	0.000	0.002	0.010	0.584	0.016	0.000	0.006	0.207
24	Fractured basement	0.000	0.338	0.006	0.001	0.000	0.017	0.008	0.008	0.001	0.010	0.172	0.000	0.007	0.000	0.000
25		0.058	0.059	0.004	0.020	0.000	0.254	0.014	0.001	0.001	0.013	0.207	0.002	0.013	0.145	0.006
26		0.351	1.759	0.018	0.025	0.000	0.263	0.015	0.033	0.005	0.013	2.531	0.016	0.036	0.000	0.012
27		0.753	0.022	0.016	0.032	0.000	1.129	0.054	0.003	0.054	0.013	0.251	0.002	0.009	0.095	0.075
28		0.265	5.423	0.101	0.009	0.007	1.257	0.063	0.003	0.007	0.005	5.140	0.061	0.022	0.071	0.255
29		0.107	0.651	0.053	0.018	0.000	0.056	0.103	0.010	0.003	0.015	0.695	0.003	0.028	0.000	0.026
30		0.165	1.054	0.037	0.024	0.000	0.197	0.112	0.007	0.003	0.012	0.796	0.005	0.018	0.018	0.213
31		0.086	0.386	0.062	0.009	0.000	0.016	0.031	0.006	0.002	0.013	1.461	0.012	0.015	0.000	0.037
32		0.062	0.763	0.036	0.009	0.000	0.000	0.006	0.040	0.004	0.009	5.514	0.028	0.013	0.000	0.000
33		0.085	0.552	0.104	0.000	0.000	0.000	0.867	0.014	0.003	0.013	2.703	0.021	0.000	0.000	0.008
34		1.246	0.755	0.070	0.005	0.003	0.619	0.208	0.001	0.007	0.005	6.687	0.034	0.012	0.011	0.191
35		0.023	2.858	0.024	0.020	0.000	0.000	0.000	0.006	0.005	0.007	11.110	0.024	0.000	0.000	0.006
36		0.072	0.631	0.046	0.007	0.000	0.000	0.001	0.002	0.005	0.004	5.475	0.022	0.016	0.004	0.031
37		0.019	1.616	0.051	0.004	0.001	0.000	0.017	0.001	0.006	0.005	10.210	0.022	0.019	0.000	0.079
Sea water		0.967	6.536	0.012	0.012	0.003	0.431	0.063	0.007	0.005	0.000	9.741	0.032	0.000	0.009	0.047
Permissible limit (EHCW 2007)		0.2	1	1	0.05	1	0.3	0.05	0.07	0.02	0.05	–	–	–	3	–

has a meteoric origin and affected by leaching and dissolution of marine or sabkha deposits as most of the groundwater samples of the alluvial, the Nubian Sandstone, and the fractured basement aquifers.

#### Rocks geochemistry

Based on field and Landsat investigations as well as to previous geochemical rock analyses, the geochemistry of both the water-bearing formation and the rocks

**Table 4** Groups of hypothetical salts for the studied groundwater samples

Aquifer	Hypothetical salts		Well numbers
	Salts	Groups	
Alluvial	NaCl, Na <sub>2</sub> SO <sub>4</sub> , MgSO <sub>4</sub> , CaSO <sub>4</sub> , Ca(HCO <sub>3</sub> ) <sub>2</sub>	IV	3, 4
	NaCl, MgCl <sub>2</sub> , MgSO <sub>4</sub> , CaSO <sub>4</sub> , Ca(HCO <sub>3</sub> ) <sub>2</sub>	V	1, 5, 6
	NaCl, MgCl <sub>2</sub> , CaCl <sub>2</sub> , CaSO <sub>4</sub> , Ca(HCO <sub>3</sub> ) <sub>2</sub>	VI	2, 7, 8, 9, 10, 11
Nubian Sandstone	NaCl, Na <sub>2</sub> SO <sub>4</sub> , MgSO <sub>4</sub> , Mg(HCO <sub>3</sub> ) <sub>2</sub> , Ca(HCO <sub>3</sub> ) <sub>2</sub>	II	14, 15, 16
	NaCl, MgCl <sub>2</sub> , MgSO <sub>4</sub> , Mg(HCO <sub>3</sub> ) <sub>2</sub> , Ca(HCO <sub>3</sub> ) <sub>2</sub>	III	17, 20
	NaCl, MgCl <sub>2</sub> , MgSO <sub>4</sub> , CaSO <sub>4</sub> , Ca(HCO <sub>3</sub> ) <sub>2</sub>	V	12, 13, 18, 19, 21, 22, 23
Fractured basement	NaCl, MgCl <sub>2</sub> , MgSO <sub>4</sub> , Mg(HCO <sub>3</sub> ) <sub>2</sub> , Ca(HCO <sub>3</sub> ) <sub>2</sub>	III	24, 25
	NaCl, Na <sub>2</sub> SO <sub>4</sub> , MgSO <sub>4</sub> , CaSO <sub>4</sub> , Ca(HCO <sub>3</sub> ) <sub>2</sub>	IV	26, 29, 30
	NaCl, MgCl <sub>2</sub> , CaCl <sub>2</sub> , CaSO <sub>4</sub> , Ca(HCO <sub>3</sub> ) <sub>2</sub>	VI	27, 28, 31, 32, 33, 34, 35, 36, 37
Rainfall	NaCl, Na <sub>2</sub> SO <sub>4</sub> , NaHCO <sub>3</sub> , Mg(HCO <sub>3</sub> ) <sub>2</sub> , Ca(HCO <sub>3</sub> ) <sub>2</sub>	I	R
Sea water	NaCl, MgCl <sub>2</sub> , MgSO <sub>4</sub> , CaSO <sub>4</sub> , Ca(HCO <sub>3</sub> ) <sub>2</sub>	V	S

along the water flow path to the wells were elucidated for each well (Table 5). In the current research, the Landsat image has provided more details about the geology which affects the recorded wells where each group of wells and in some cases each well have a different geologic setting. The details about geochemistry of rocks which interact directly with groundwater (water-bearing formation) and other rocks which interact with surface water and recharge the aquifers are concluded in (Table 5). The alluvial aquifer is composed of sand, mud, and basement rock fragments in gravel and sand sizes. The geology of watersheds along the path to the alluvial wells has the following great varieties: older gray granite (well no. 1, Fig. 4a), metamorphic rocks (gneiss, amphibolites, schist; well no. 2, Fig. 4a), metagabbro, metasediments (well nos. 3 and 4, Fig. 4b), clay, sand and evaporates (well nos. 5 and 6, Fig. 4c), metasediments, ophiolitic basic metavolcanics, and Tertiary olivine basalt (well nos. 7 to 11, Fig. 4d). The recorded wells and springs which are penetrating the Nubian Sandstone Aquifer show a homogeneity in the lithologic nature where both the water-bearing and watersheds composed of sandstone with iron-ore beds, shale, and silt intercalation and coarse to medium fluvialite sandstone (well nos. 12 to 23, Fig. 5a, b). The mineralogy of these ferruginous sandstone and ironstone includes goethite, magnetite, quartz, illite, halite, kaolinite, dolomite, and vesuvianite with abundant SiO<sub>2</sub>, Fe<sub>2</sub>O<sub>3</sub>, Al<sub>2</sub>O<sub>3</sub>, CaO, and MgO. On the other hand, the basement aquifer was explored through 15 wells penetrated five water-bearing rocks as follows: ophiolitic metagabbro (well nos. 24 and 25, Fig. 6c), older gray granite (well no. 26, Fig. 6b), metasediments (well no. 27, Fig. 6b), acid volcanic (well nos. 28 to 35, Fig. 6a), and ophiolitic serpentinite with talc carbonate (well nos. 36 and 37, Fig. 6d), (Table 5).

### Thematic layers affecting groundwater

The data obtained from the investigation of Landsat images and DEM analyses were used for the generation of the thematic layers which are affecting the groundwater occurrences and quality and are shown in Figs. 7, 8, 9, 10, 11, and 12.

### Structural lineaments

The structural lineaments of the area under investigation were elucidated and stated this area was affected by sets of faults where the trends of lineaments are NE-SW, E-W, and some NW-SE which are parallel to the Red Sea Rift (Fig. 7). These lineaments are concentrated in the southern and western parts of the study area where Nubian sediments and basement rocks are located. The Nubian Sandstone is fractured to a lesser degree than the basement rocks; these fractures are playing the main role in both groundwater occurrences and its recharge. With the comparison between drainage networks layer (Fig. 10) and structural lineament layer (Fig. 7), it is assumed that drainage lines are structurally controlled and developed along faults and/or fractures.

### Elevation

The elevation difference over the study area reaches about 1700 m which reveals a great variation in the terrain shape. The coastal plain shows the lower elevation areas as well as the areas along the drainage lines in between the Red Sea Mountains. These lowlands are representing the water collectors where the majority of surface water running and can enhance the groundwater recharge (Fig. 8), while the high lands represent the main watersheds which receive any rainfalls and directed to the lower-altitude areas.

### Drainage networks and hydrographic basins

The hydrographic basins were extracted through DEM analyses where the area has about 16 main basins (Fig. 9). These basins were named according to topographic maps,

**Table 5** Geology and geochemistry of water bearing formation and watershed affecting groundwater wells

Well no.	Aquifer	Water-bearing formation	
		Lithology	Geochemistry and/or mineralogy
1	Alluvial	Alluvial sediments composed of sand, mud, and basement rock fragments in gravel and sand sizes	
2		Rock fragments of basement and mixture of watershed rocks; their geochemistry are identified in the following columns. In addition, carbonate minerals are dolomite, calcite, Mg-calcite, and aragonite. Siliciclastic minerals are quartz, K-feldspars, and plagioclase, with minor heavy minerals and clays (SiO <sub>2</sub> , CaO, Al <sub>2</sub> O <sub>3</sub> , and MgO)	
3			
4			
5			
6		Alluvial sediments and sabkha deposits	Evaporates minerals include (gypsum, halite, anhydrite). Carbonate minerals are dolomite, calcite, Mg-calcite, and aragonite. Siliciclastic minerals are quartz, K-feldspars, and plagioclase (SiO <sub>2</sub> , CaO, SO <sub>3</sub> , MgO and Al <sub>2</sub> O <sub>3</sub> )
7 to 11		Alluvial sediments composed of sand, mud, and basement rock fragments	The same as mentioned in well numbers 1 and 2
12 to 19	Nubian Sandstone	Sandstone with iron-ore beds, shale, and silt intercalation	
20 to 23		Coarse to medium fluvial sandstone	
24	Fractured basement	Ophiolitic metagabbro	
25		Older gray granite	
26			
27			
28 to 35			
36		Metasediments	
37	Acid volcanics		Include rhyolite, rhyodacite, and dacite. High concentrations of SiO <sub>2</sub> , Na <sub>2</sub> O, Fe <sub>2</sub> O <sub>3</sub> , K <sub>2</sub> O, Zr, Nb, and Y but low in MgO, CaO, Cr, Ni, Sr, Ga, and V
36		Ophiolitic serpentinite and talc carbonate	
37		Include magnizite, chromite, dolomite, talc, iron oxides (goethite, hematite, magnetite). Abundant Al <sub>2</sub> O <sub>3</sub> , SiO <sub>2</sub> , FeO, TiO <sub>2</sub> , CaMg(CO <sub>3</sub> ) <sub>2</sub> , CaCO <sub>3</sub> , and Fe-Ti oxides	
Well no.	Aquifer	The geology along the water flow path to the wells	
		Lithology	Geochemistry and/or mineralogy
1	Alluvial	Older gray granite	
2		Metamorphic rocks	
3		Metagabbro, metasediments	
4		Sabkha deposits included clay, sand, and evaporates	
5			
6			
7 to 11			
12 to 23	Nubian Sandstone	Metasediments, ophiolitic basic metavolcanics, Tertiary olivine basalt	
		Sandstone with iron-ore beds, shale, and silt intercalation	
		Metasediments: high Al <sub>2</sub> O <sub>3</sub> , CaO, Fe <sub>2</sub> O <sub>3</sub> , MgO, TiO <sub>2</sub> , Ni, V, and Cr; low SiO <sub>2</sub> , Na <sub>2</sub> O, Rb, Ba, Sr, Zr, and Y, where Tertiary olivine basalt composed of plagioclase, augite, and olivine (SiO <sub>2</sub> , TiO <sub>2</sub> , K <sub>2</sub> O, MgO) (R1, R3)	
		Ferruginous sandstone and ironstone. The mineralogy includes goethite magnetite, quartz, illite, halite, kaolinite, dolomite, and vesuvianite. Abundant SiO <sub>2</sub> , Fe <sub>2</sub> O <sub>3</sub> , Al <sub>2</sub> O <sub>3</sub> , CaO, MgO (R4)	

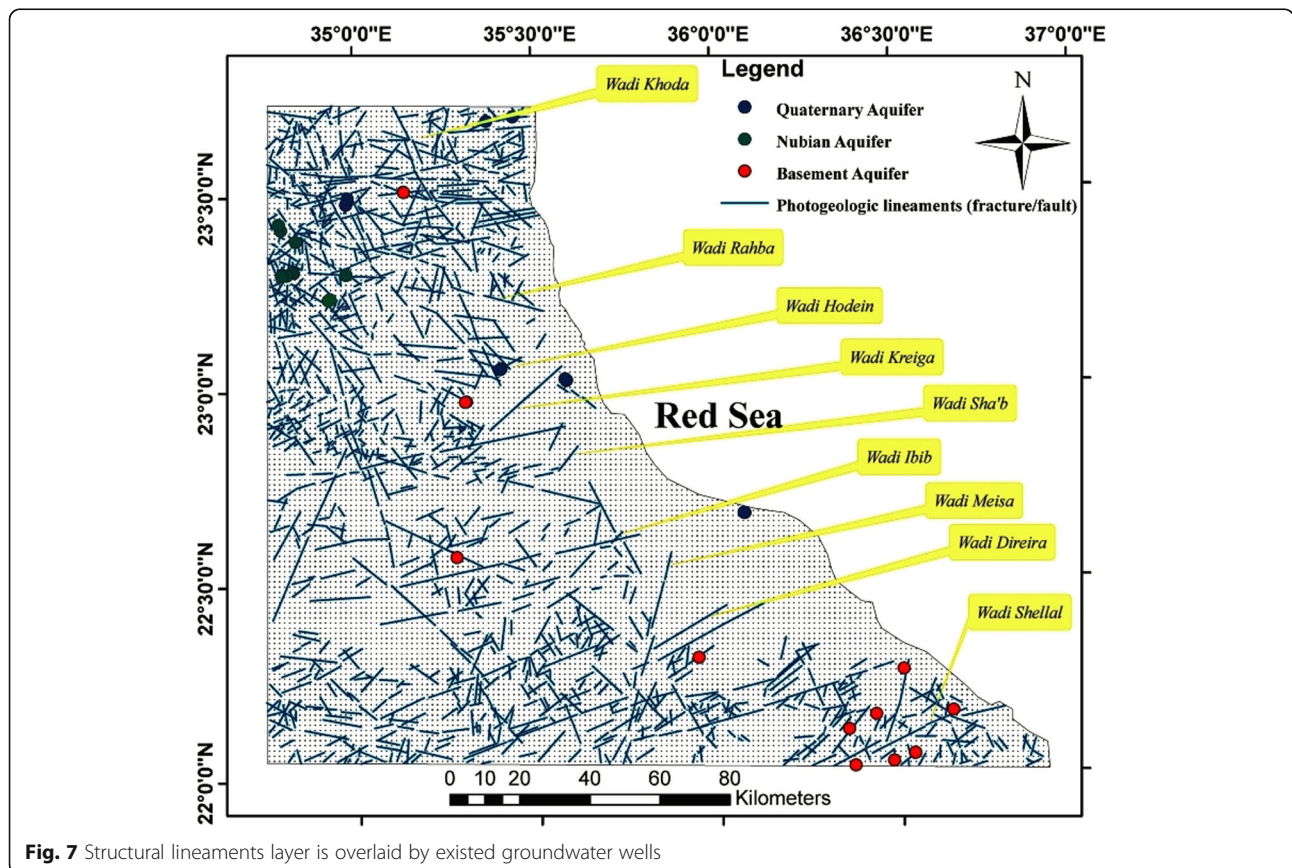
**Table 5** Geology and geochemistry of water bearing formation and watershed affecting groundwater wells (*Continued*)

Well no.	Aquifer	Water-bearing formation	
		Lithology	Geochemistry and/or mineralogy
24	Fractured basement	Ophiolitic serpentine and talc carbonate	Include magnizite, chromite, dolomite, talc, and iron oxides (geothite, hematite, magnetite). Abundant Al <sub>2</sub> O <sub>3</sub> , SiO <sub>2</sub> , Feo, TiO <sub>2</sub> , CaMg(Co <sub>3</sub> ) <sub>2</sub> , CaCO <sub>3</sub> , and Fe-Ti oxides (R1)
25			
26	Acid volcanic		Include rhyolite, rhyodacite, and dacite. High concentrations of SiO <sub>2</sub> , Na <sub>2</sub> O, Fe <sub>2</sub> O <sub>3</sub> , K <sub>2</sub> O, Zr, Nb, and Y but low in MgO, CaO, Cr, Ni, Sr, Ga, and V (R5)
27	Metamorphic rocks		The same like well number 2
28 to 35	Acid volcanic and metasediments		Acid volcanic (the same as well number 26) Metasediments (the same as well number 4)
36	Older gray granite and acid volcanic		Older gray granite (the same as well number 1)
37			Acid volcanic (the same as well number 26)

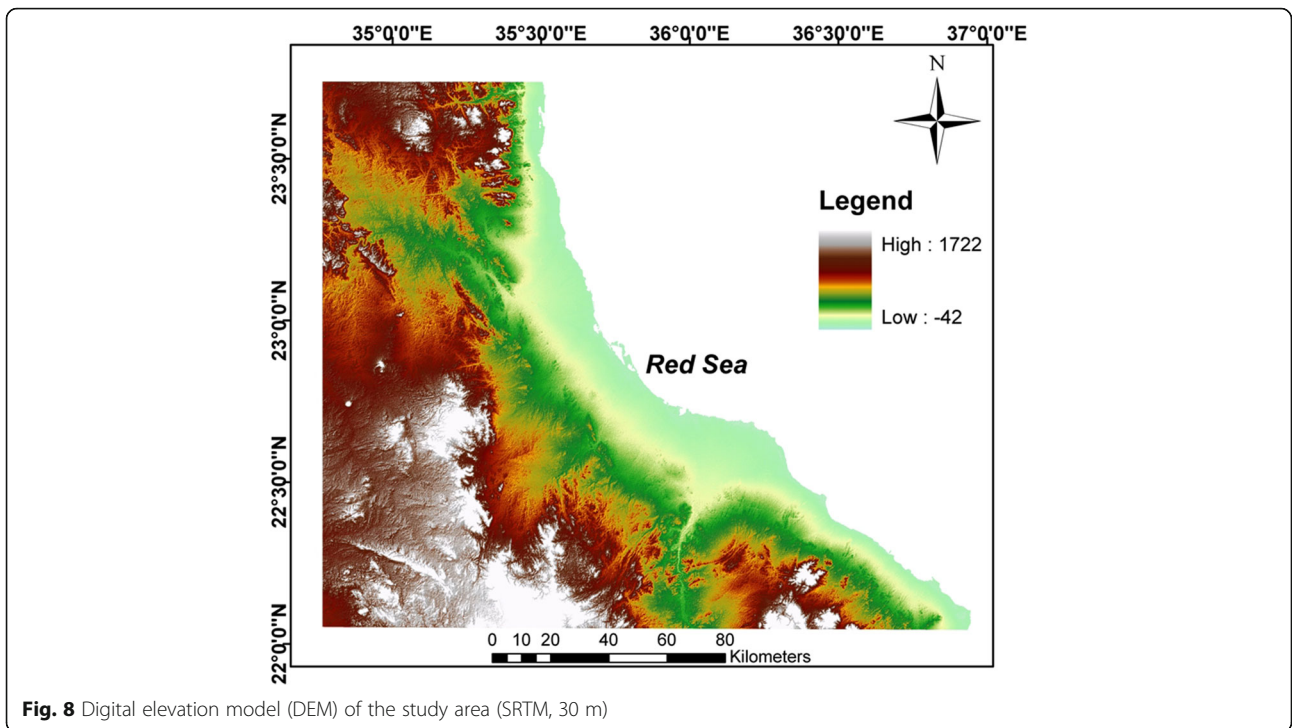
The geochemistry of the studied rocks is obtained from different sources as follows: (R1)—Hassan (2015), (R2)—Abd El-Naby and Frisch (2006), (R3)—Khalil (1997), (R4)—Farouk et al. (2015), (R5)—Shahin et al. (2016)

and their names were shown in (Figs. 1 and 7). Only eight basins are containing groundwater wells; the Nubian Sandstone Aquifer is concentrated in the Wadi Hodein Basin, while the wells of alluvial and basement aquifers are scattered inside other basins. These basins are representing the catchment areas affecting the existing groundwater wells. The hydrographic basins include high intensive drainage network (Fig. 10) which provides more chances for

groundwater recharge from surface runoff. These drainage lines are controlled by fractures and/or faults which mean that the sites along the drainage lines are associated by structural lineaments. This condition facilitates chances for groundwater accumulation which can be recharged through surface-groundwater interactions. It is also noted that the recorded well were drilled along the drainage lines except well nos. 5 and 6 from the alluvial aquifer.



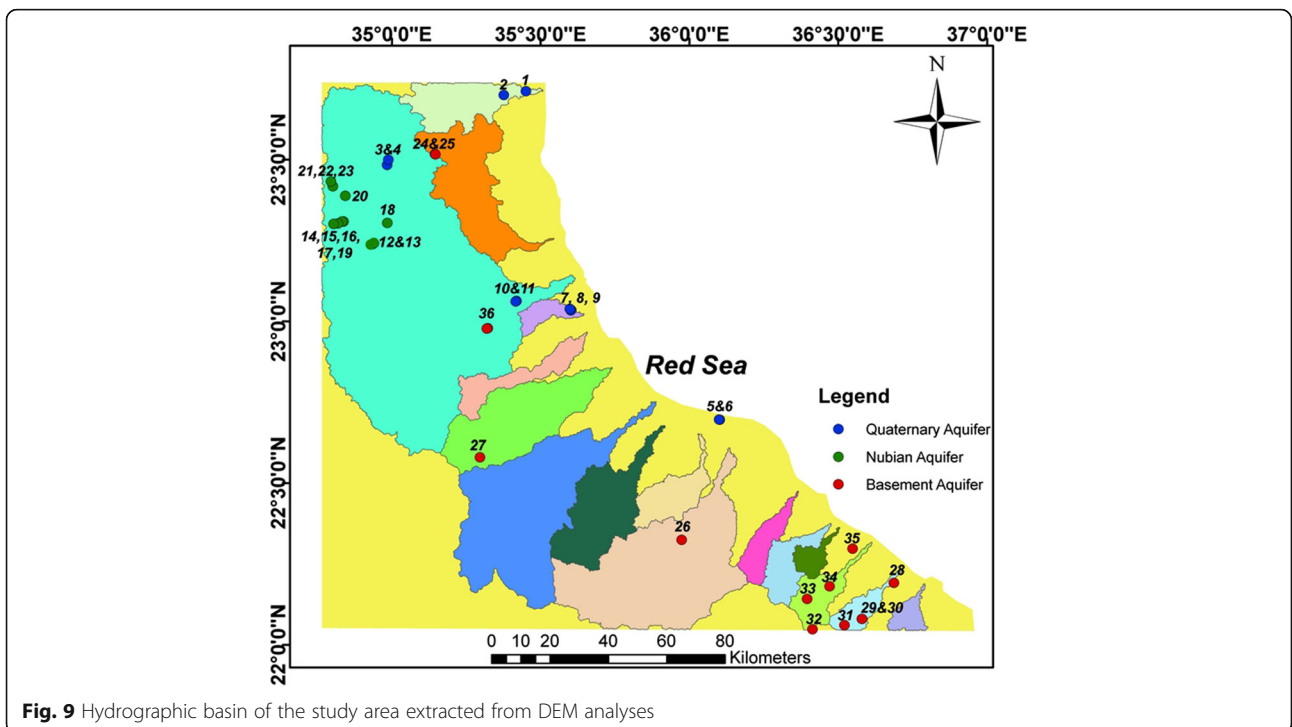
**Fig. 7** Structural lineaments layer is overlaid by existed groundwater wells

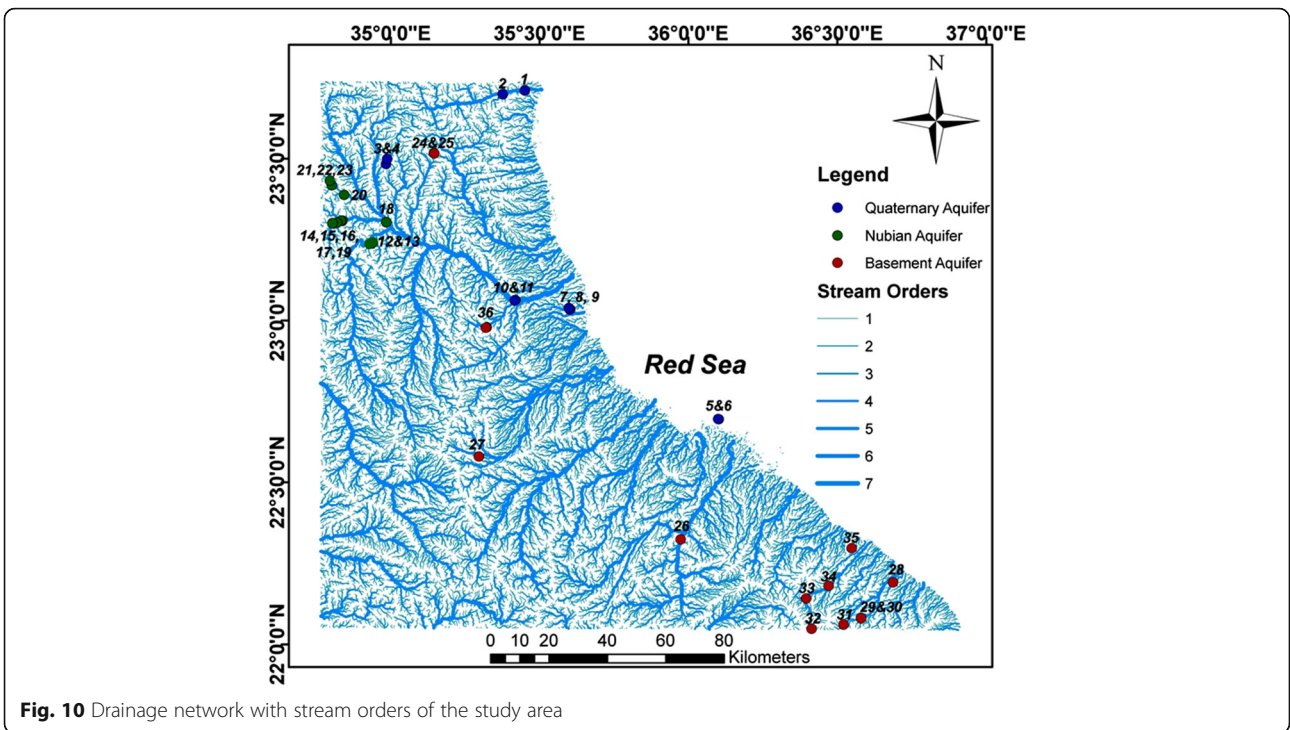


**Flow direction**

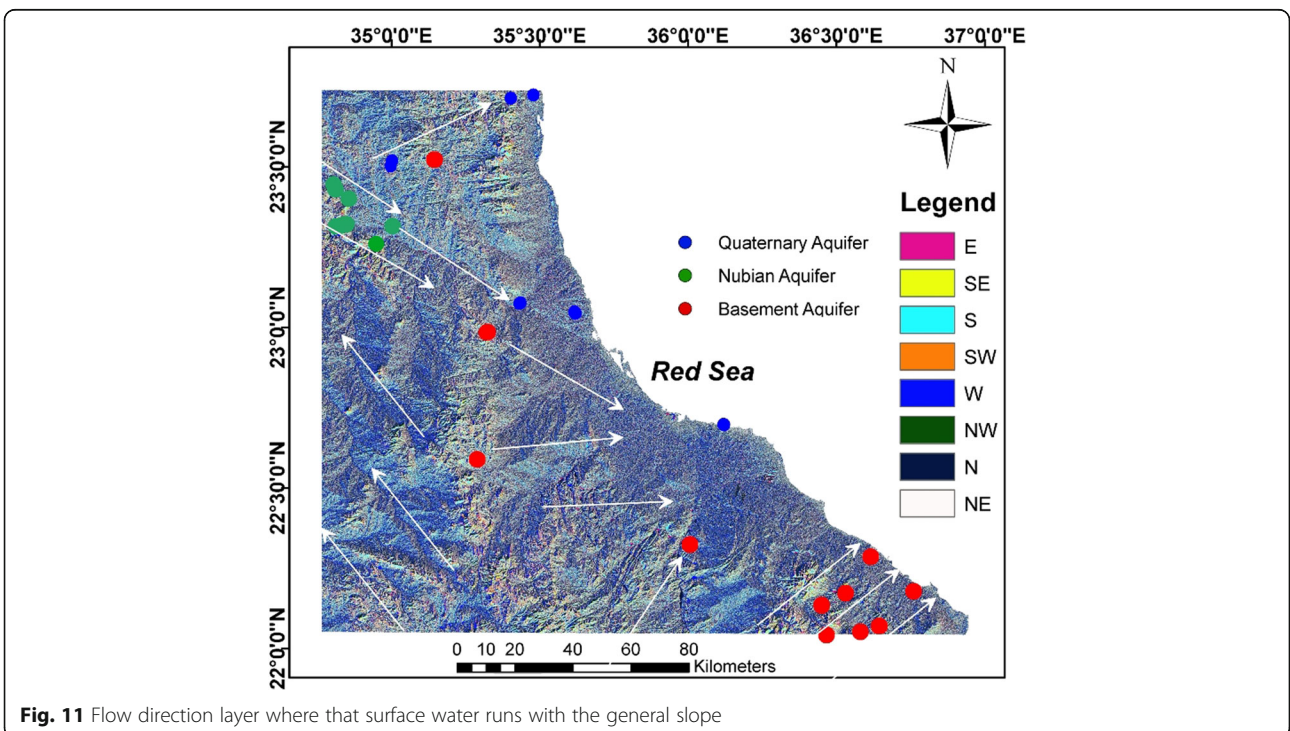
Flow direction is one of the important layers which contribute to the assessment of groundwater (Fig. 11). The layer of flow direction shows that surface water runs with the general slope with some exception for the isolated hilly areas

where they direct surface water to the adjacent lower areas. The common flow directions have trends to the northeast and southeast. It is noted that the recorded groundwater wells are located in the path of surface flow direction which supported the hypothesis of the current research.



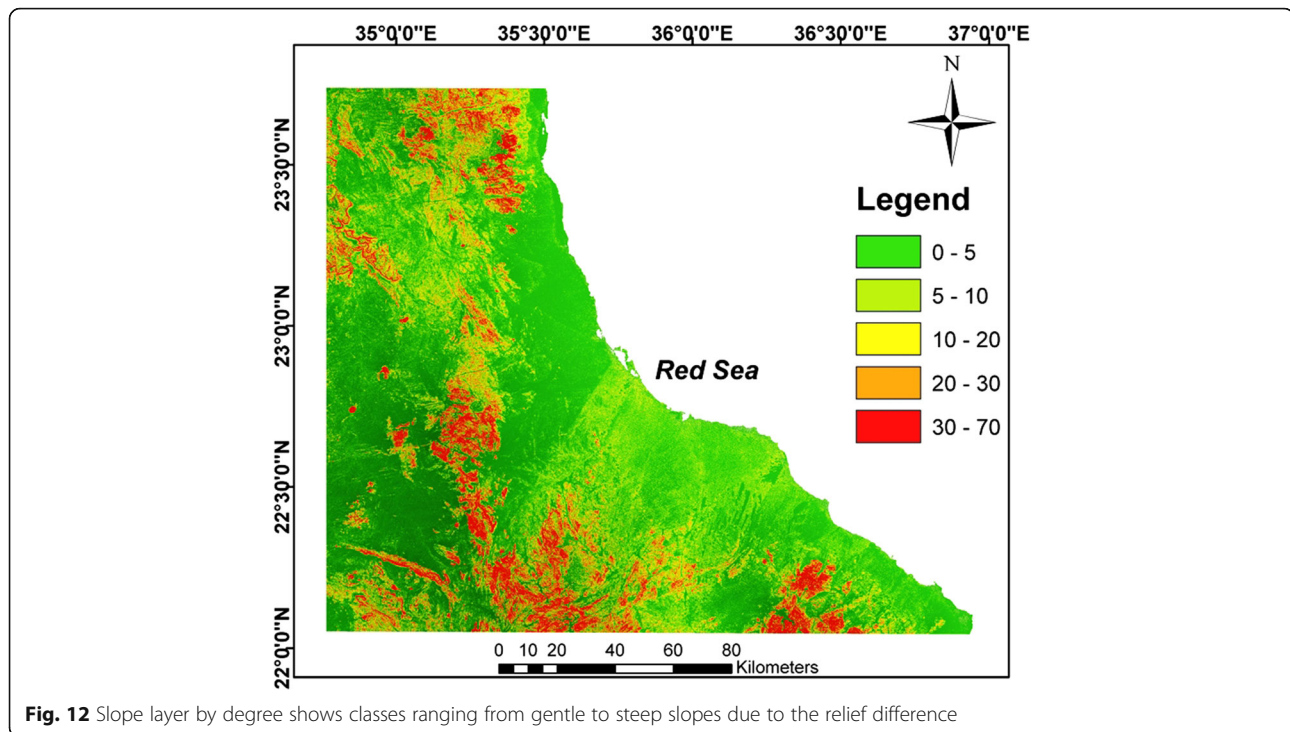


**Fig. 10** Drainage network with stream orders of the study area



**Fig. 11** Flow direction layer where that surface water runs with the general slope





### Slope

The slope layer shows (Fig. 12) the values vary between 0 and 10° for the coastal plain and along the drainage lines and 20–70 for the Red Sea terrains and sedimentary blocks. The Red Sea terrains have steep slopes due to the relief difference while the coastal plain and drainage networks are acting as recharge areas.

### Discussion

The chemical composition of groundwater is a combination of water constituents that migrate into the groundwater reservoir with recharge waters and a reaction of the groundwater body with the minerals present in the aquifer matrices (Yousif et al. 2016). The practice of environmental sustainability aims to secure the necessary requirements of the population without deterioration of the ability of the natural resources (such as groundwater) to meet the needs of next generations. Here, the current research is focusing on the environmental protection from the side of groundwater through the understanding of rock-water interaction.

### Rock-groundwater interaction and speciation modeling

In general, multi-chemical processes occur during the rock-water interaction, such as dissolution and/or precipitation, but it is worthy to mention that other factors can also contribute to the groundwater chemistry like recharge amounts, seawater intrusion, and any source of pollution. Here, the research is focusing on the chemical composition of groundwater which is influenced by rock interaction.

Interpretation of chemical data suggests that the dissolution of  $\text{CaCO}_3$ , ion-exchange processes, and weathering of minerals control the major ion chemistry of sodium, calcium, magnesium, and potassium. With the comparison between results in Tables 4, 5, and 6, it is clear that groundwater chemistry is guided by rocks influences where dissolution of minerals and accessory minerals are the main sources for major and trace elements. Taking into account, the study area is subjected to some rainy storms (the last heavy storm was recorded on 19 May 2017 where the rainfall continue for 2 h and resulted in surface runoff, this data was recorded through fieldwork) which lead to the recharging of the groundwater. The isotope analyses (oxygen 18 and deuterium) which carried out by Elewa (2000) confirmed the occurring of this recharging process. The freshwater supply to this area was depending on some natural springs from Nubian Sandstone exposures where the past historical reports mentioned that one of the French explorers (Linant de Bellefonds) recorded spring beside a temple cut into a mountainside that he saw in March 1832 in the southern part of Egypt's Eastern Desert (Sidebotham et al. 2004). This spring is still providing freshwater until the present which reveals the continue recharges to the Nubian sandstone aquifer. Therefore, the leaching and dissolution can be occurring through the interaction between the surface water and watershed rocks and also between water-bearing formation and groundwater. It is worth to mention that the recharge of the Nubian Sandstone Aquifer is a process which is controlled by some factors. For example, in the Western Desert, recharge

**Table 6** Saturation indices of the studied groundwater samples, using PHREEQC geochemical model

Well no.	Aquifer	Aragonite, CaCO <sub>3</sub>	Calcite, CaCO <sub>3</sub>	Dolomite, CaMg(CO <sub>3</sub> ) <sub>2</sub>	Alunite, KAl <sub>3</sub> (SO <sub>4</sub> ) <sub>2</sub> (OH) <sub>6</sub>	Gibbsite, Al(OH) <sub>3</sub>	Goethite, FeOOH	Hematite, Fe <sub>2</sub> O <sub>3</sub>	Iron(III) oxide- hydroxide	Magnetite, MnOOH	Barite, BaSO <sub>4</sub>
1	Alluvial	0	0.09	0.31		0.82	7	17		13	
2		0	0	0	6	3	9	21			
3		0.61	0.75	1.75		1.3	7	18			
4		0.66	0.8	2							
5		0	0.02	0.14	2	1.65	7	18		13	0.4
6		0.24	0.39	0.89	0.35	1.44	7	18		14	0.35
7					4	8	2.59	20		13	
8		0	0.08	0.1	3.44	2.32	8.49	20		13	
9					3.14	2.12	8.16	19.32		12.76	
10		0	0.14		1.68	1.31				13.9	
11		0	0.07		1.14	1.44	13.8				
12	Nubian Sandstone					1	9	21	2	10	
13						2	9	21	4	20	
14						1.16	9.19	21.38	2.3	17.15	
15						1.77	9.51	22	2.6	17.5	
16						0.86	9.5	22	2.6	16.15	
17						1.44	8.24	19.48	1.34	17.6	
18						1.4	8.34	19.69	1.45	14.94	
19						1.42	7.92	18.84	1	14.79	
20						1.4	7	18	1	16.19	
21						1.4	7.92	17	1	15.67	
22						1.37	7.76	18.52	0.9	15.66	
23						1.4	7.53	18.07	0.7	13.82	
24	Fractured basement			0.58				17.52			
25				0.53	0	0.67	8.45	19.9			
26		0	0.11	0.33	0.48	1.79	8.33	19.66		16	
27				0.06	0	2	9	21		16.27	
28					1.89	1.61	7.95	18.9		11.12	
29		0	0.05	0.09	0	1.31	7.62	18.25			
30		0	0.01		0	1.49	8.22	19.45		16.16	
31		0.19	0.34	0.44	0	1.48	6.98	16.97		14.7	
32		0.22	0.37	0.68	1.7	1.64				12.43	0.1
33		0	0.14	0.07	1.46	1.8				14.79	0.25
34					5.3	2.78	8.09	19.2		13.33	0
35		0.04	0.19	0.09	1.47	1.19					0
36		0.04	0.18	0.24	0.62	1.41				13.43	0
37		0.49	0.63	1.21	0.52	1.03				13.26	0.02

The + values of the saturation indices express the over-saturated phase, the 0 values of the saturation indices express the saturated phase

of this aquifer can happen under extreme precipitation over the Nubian watershed in Sudan and Chad or under a considerable rising of Lake Nasser levels (Ahmed and Abdelmohsen 2018).

The groundwater chemistry of this area is principally controlled by the mixing of waters, evaporation, mineral

equilibrium, dissolution and deposition, and ion exchange processes (Dehnavi et al. 2011). The obtained results of saturation indices (SI) of minerals are useful for evaluating the level to which chemistry of water is controlled by the equilibrium with solid phases (Appelo and Postma 1993). The saturation indices (SI) of minerals that were expected

to be responsible for the chemical composition of the recorded groundwater were investigated.

The obtained results of speciation modeling are summarized in (Table 6) where samples from the three studied aquifers were investigated. The three aquifers are supersaturated with respect to gibbsite ( $\text{Al}(\text{OH})_3$ ), goethite ( $\text{FeOOH}$ ), hematite ( $\text{Fe}_2\text{O}_3$ ), and magnetite ( $\text{MnOOH}$ ). All samples are supersaturated with gibbsite mineral due to the presence of aluminum in suspended fraction where the geochemistry of water-bearing formations and the rocks along the water path (Table 5) shows that  $\text{Al}_2\text{O}_3$  is a common component. The investigation of Landsat images and geologic map with respect to the locations of groundwater wells (Figs. 3, 4, 5, and 6) shows the geology along the water path to these wells which composed of metasediments, metamorphic, and igneous rocks in case of alluvial and basement aquifers. The geochemistry of these rocks (Table 5) stated that  $\text{Fe}_2\text{O}_3$  and  $\text{MgO}$  are abundant minerals. Therefore, their groundwaters are supersaturated with goethite, hematite, and magnetite. On the other hand, the outputs of speciation modeling with regard to Nubian Sandstone Aquifer showing homogeneity results where its groundwater samples are supersaturated with gibbsite, goethite, hematite, magnetite, and iron(III) oxide-hydroxide. These supersaturations and homogeneity in mineral indices resulted from ferruginous sandstone and ironstone which represent the water-bearing formation and also the main watershed which reacts with the surface water through its path to recharge groundwater. The other results of geochemical model clarify that groundwaters from both alluvial and basement aquifers are supersaturated with aragonite ( $\text{CaCO}_3$ ), calcite ( $\text{CaCO}_3$ ), dolomite ( $\text{CaMg}(\text{CO}_3)_2$ ), and alunite ( $\text{KAl}_3(\text{SO}_4)_2(\text{OH})_6$ ) (Table 6). In case of alluvial aquifer, the water-bearing formation consists of rock fragments of basement rocks and a mixture of carbonate minerals such as dolomite, calcite, Mg-calcite, and aragonite. Siliciclastic minerals are quartz, K-feldspars, and plagioclase, with minor heavy minerals and clays ( $\text{SiO}_2$ ,  $\text{CaO}$ ,  $\text{Al}_2\text{O}_3$ , and  $\text{MgO}$ ). On the other side, the basement aquifer has a variety of rock content which includes acid volcanic, granite, ophiolitic metagabbro, and talc carbonate. These rocks are rich with  $\text{CaMg}(\text{CO}_3)_2$ ,  $\text{CaCO}_3$ ,  $\text{CaO}$ , and  $\text{K}_2\text{O}$  which can be considered the main source for the saturation with carbonate minerals in groundwaters. However, rocks are generally classified as sedimentary, igneous, and metamorphic where minerals formed these rocks can be dissolved in the groundwater based on their resistance to the chemical weathering. These reactions between rocks and groundwater are depending on the initial water which in this case is represented by rainfall. Therefore, the current research conformed that the groundwater chemistry is the imprints of the rock-water interaction.

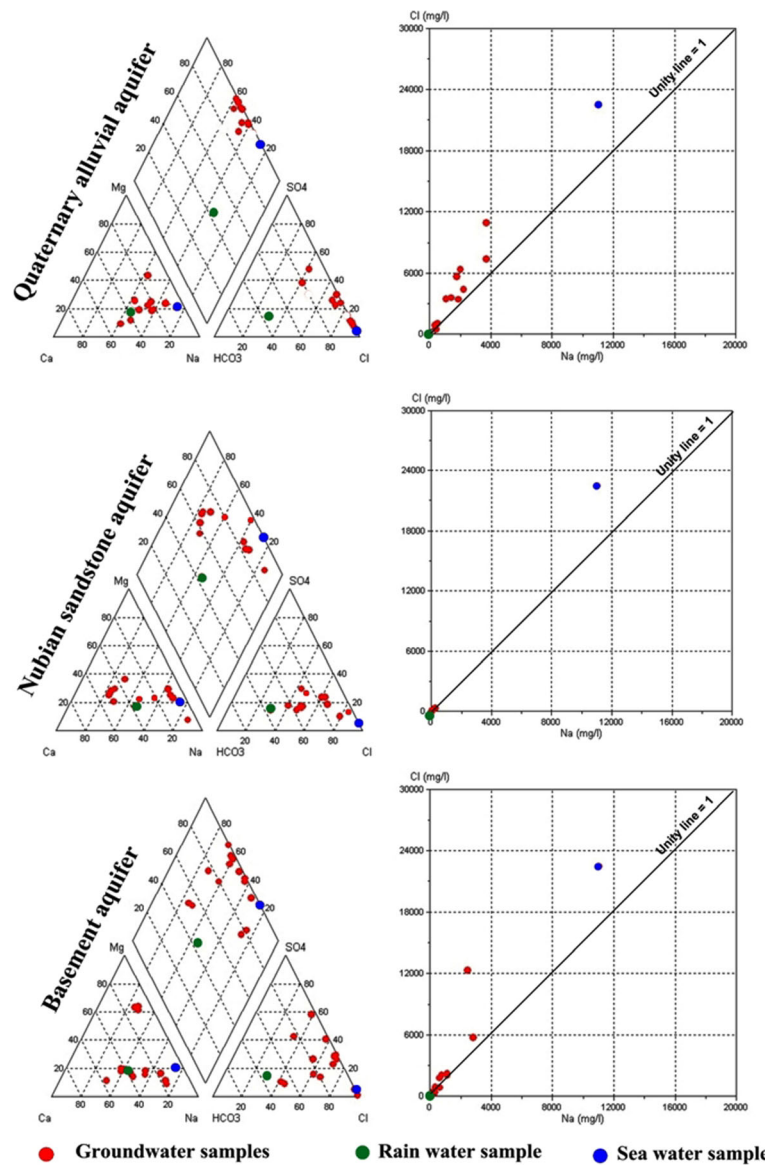
### Hydrogeochemical evolution and groundwater genesis

Piper's diagram (Piper 1944) in (Fig. 13) consists of three separate fields for plotting chemical data, two triangles at the lower right and lower left sides, and an intervening diamond shape for the projection. The major cations ( $\text{Ca}^{2+}$ ,  $\text{Mg}^{2+}$ , and  $\text{Na}^+$  plus  $\text{K}^+$ ) are plotted in the lower left triangle, while the major anions ( $\text{Cl}^-$ ,  $\text{SO}_4^{2-}$ , and  $\text{CO}_3^{2-}$  plus  $\text{HCO}_3^-$ ) are plotted in the lower right triangle. The two points representing the cations and anions composition are then projected on the diamond-shaped field. Distinct groundwater quality types can be easily distinguished by their plotting in certain sub-areas of the diamond-shaped field. Piper's diagram suggested that the groundwater in the studied area have the meteoric origin of mineralization (precipitation and flash flood over the area) affected by leaching and dissolution, this is for few samples of the investigated aquifers of the study area. The ion dominance of this groundwater, especially in the Nubian sandstone aquifer, possessed the sequence of  $\text{Na} > \text{Ca} > \text{Mg}$  with  $\text{Cl} > \text{HCO}_3 > \text{SO}_4$ . Most groundwater samples of alluvial, Nubian Sandstone, and fractured basement aquifers are affected by leaching and dissolution of marine deposits and considered mature water.

### Groundwater protection and development (recommendation)

In arid and hyper-arid regions, groundwater can be considered the sole source of water for domestic, industrial, and agricultural needs. The development and protection of this precious source are a prerequisite for both potential contamination and overexploitation. Therefore, the understanding of the hydrogeologic and geologic settings as well as the factors which affect the groundwater occurrences and quality are in dire needs. Knowledge of aquifer lithology and identification of geology along the flow path to the groundwater wells, in addition to investigations of different thematic layers related to groundwater accumulation, was used to summarize some recommendations. Based on field investigations, remote sensing data, geologic situation, geomorphologic setting, hydrologic parameters of the recorded well data, groundwater chemistry, and the thematic layers (structural lineaments, DEM, hydrographic basins, drainage lines, flow direction, and slope), the following suggestion can be applied:

1. Priority map for groundwater exploration is proposed for both hydro-geophysical investigation and test drillings in order to evaluate the potential of groundwater (Fig. 14). This map was created according to the results of the multidisciplinary data layer analysis of this study by using Arc GIS software. It is expected that the very high and high priorities have a good potential for groundwater

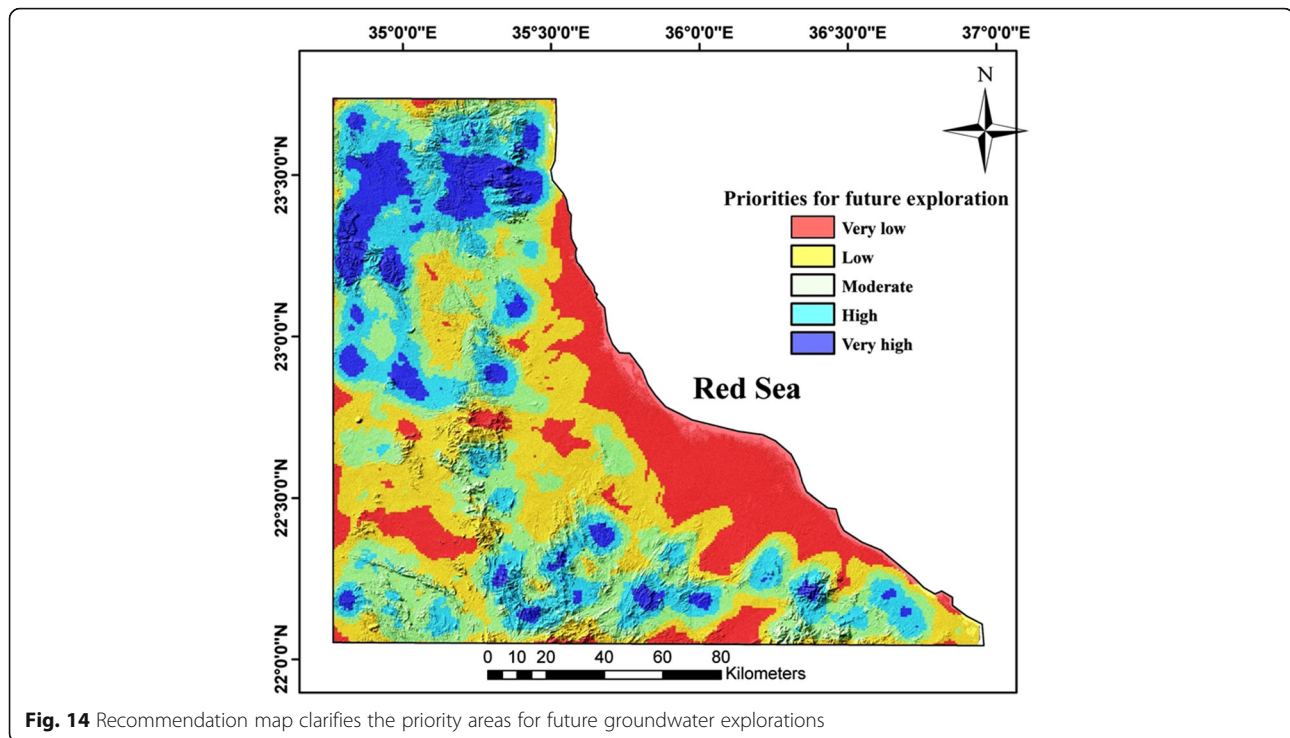


**Fig. 13** Groundwater wells from the studied aquifers plotted on Piper diagram and scattered plot of Na-Cl

recharge based on the criteria such as density of structural lineaments (faults and/or fractures) which provide more chance for groundwater recharge, low slope, permeable soil type, position along the drainage network, elevation, and flow direction. The drainage lines and flow direction also support the chances for recharge. The low elevation and low slope values are increasing the residence time of surface water and hence providing good conditions for water accumulation. The increasing of residence time in fractured areas which have a low slope and elevations raise the possibility of recharge processes. In case of basement aquifer, the design for determination of the priorities for exploring

groundwater was depending mainly on careful investigation of the structural lineaments with regard to their orientation, and the density of the intrusive dykes where it acts as a barrier of the percolated water. The intersected fractures with each other's and intrusive dykes are providing high possibility environments for groundwater entrapments. Therefore, these areas are suggested for future drilling.

2. The coastal plain and some parts of Piedmont Plain as wells as intermountain areas have low and very low priorities (Fig. 14) due to low fracture density and the possibility of seawater intrusions which can be increased with aquifer discharging.



- The obtained data about the recorded wells were used to calibrate the priority map (Fig. 14). With regard to TDS values, the wells which are producing a freshwater (such as, well nos. 12, 14, 15, 16, 17, 18, 19, 20, 21, 22, 23, 24, 25, 27, 29; Table 1) are located in the areas with priority classes from high to very high (compare Fig. 10 “drainage with wells locations” with Fig. 14), which confirm the basics for creations of the priority map.
- The obtained results showed that the people in the study area are suffering from water shortage problem, and the available water resources are polluted by heavy metals and/or maybe microbial pollutants. The trace element concentration of the investigated groundwater samples showed some values exceed the permissible limit (according to EHCW 2007) which represent a health hazard in case of drinking that water. We also noticed a thick layer of green algae covered the pool that collected water from Abu Safa spring (Fig. 1c) of the Nubian Sandstone Aquifer, which can be considered a source of pollution. Therefore, we suggest that the government can install some water treatment plants especially near the polluted freshwater resources (Wadi Abu Safa) to remediate the pollution in these water springs. Also, we suggest installing photoreactors that can be used in this remote area for water purification and disinfection via the photo-catalysis technique as one of the advanced oxidation processes without any byproduct.
- The best solution to overcome the saline and brackish water is the installation of the desalination plant. In such cases, the desalination of this groundwater provides coast reduction than desalination of seawater. Desert Research Center and Egypt Desalination Research Center of excellence have the required experiences to install small mobile water desalination plants using solar power. These small plants are considered the suitable solution for the high salinity problem in the study area.
- Well nos. 12 and 13 of Nubian Sandstone Aquifer are very adjacent to each other, where the salinity values show the wide difference (TDS for well 12 is 776 mg/l and for well 13 is 14,388 mg/l). The review of the drilling reports of both wells stated that after 120 m (in case of well no. 12), the sandstone was completely affected by hydrothermal solutions which maybe came out through the fracture. This solution is very rich with minerals and salts and led to lost or reduces the porosity and permeability and also increase the salinity of groundwater (as in case of well 13). Therefore, it is recommended to avoid installation of screens in front of any sandstone affected by hydrothermal solutions during the design of any wells penetrate the Nubian Sandstone Aquifer. Shallow depths between 90 and 120 m can provide good potential and quality of groundwater, taking into account cost reduction.

## Conclusions

This study presents an example from a hyper-arid area on the Eastern Desert of Egypt where a combination of multidiscipline was applied to describe the rock-water interaction by testing the ability of remote sensing data to provide details about the geology affecting each groundwater well. In general, multi-chemical processes occur during the rock-water interaction, such as dissolution and/or precipitation, but other factors can also contribute to the groundwater chemistry like recharge amounts, seawater intrusion, and any source of pollution. Here, the research is focusing on the chemical composition of groundwater which is influenced by rock interaction. Interpretation of chemical data suggests that the dissolution of  $\text{CaCO}_3$ , ion exchange processes, and weathering of minerals controls the major ion chemistry of sodium, calcium, magnesium, and potassium. However, rocks are generally classified as sedimentary, igneous, and metamorphic where minerals formed these rocks can be dissolved in the groundwater based on their resistance to the chemical weathering. These reactions between rocks and groundwater are depending on the initial rainfall; therefore, the current research conformed that the groundwater chemistry is the signature of the rock-water interaction. The leaching and dissolution can occur through the interaction between the surface water and watershed rocks and also between water-bearing formation and groundwater. The integration of remote sensing and GIS application is showing great efficiency in the knowledge of lithologic nature and identification of geology along the flow path to the groundwater wells. Also, these applications provide different thematic layers related to groundwater accumulation (geomorphologic setting, hydrologic parameters, groundwater chemistry, and the thematic layers such as structural lineaments, DEM, hydrographic basins, drainage lines, flow direction, and slope) which can be used for the determination of the priorities for exploring groundwater. The current scientific approach can be considered a replicable model which can be applied in the hyper-arid environment under the same conditions.

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## Availability of data and materials

All data generated or analyzed during this study are included in this manuscript.

## Authors' contributions

MY studied the geology of the study area, analyzed and interpreted different satellite images and digital elevation models, prepared the maps through the applications of GIS, and run the geochemical model. AHE-A performed the chemical analyses of the groundwater, collected the groundwater samples, and calculated the hypothetical salts. Both authors wrote, reviewed, and approved the final manuscript.

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## Ethics approval and consent to participate

Not applicable

## Consent for publication

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## Competing interests

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